

## Structural and Luminescent Properties of Samarium Doped SrSnO<sub>3</sub> Nanoparticles

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The alkaline earth stannates (ASnO<sub>3</sub>) have been recently investigated for their potential applications, such as components in lithium ion batteries and high-temperature humidity sensors [1]. Rare earth element doped semiconductors materials and perovskites, such SrSnO<sub>3</sub>, are of interest for visible and infrared light emitting diodes applications [2]. Among the rare earth elements, samarium (Sm<sup>3+</sup>) is an element which is often employed as dopant for making orange-emitting phosphors [3]. In this work, we present the synthesis of SrSnO<sub>3</sub>:Sm<sup>3+</sup> nanoparticles using a co-precipitation method and investigated their structural properties and photoluminescent characteristics.

In a typical co-precipitation synthesis [4], 55 mL of a 0.02M solution of SnCl<sub>4</sub>•5H<sub>2</sub>O and 55 mL of a 0.055M solution of Sr(NO<sub>3</sub>)<sub>2</sub>•6H<sub>2</sub>O were prepared in deionized water. 60 mL of hydrogen peroxide was added independently in both solutions. While keeping the SrNO<sub>3</sub> solution under constant stirring, 11 ml of ammonium hydroxide were added to this solution. After 30 min, the SrNO<sub>3</sub> solution was slowly mixed with SnCl<sub>4</sub> solution, acquiring a cloudy yellow appearance. For Sm-doped samples, Sm(NO<sub>3</sub>)<sub>3</sub>•6H<sub>2</sub>O was added in the following molar concentrations (1, 2 and 4 %). Finally, the samples were washed and centrifuged several times alternating between water and ethanol. Obtained precipitates were dried in an oven at 80 °C for 2 h. Samples were sintered at two different temperatures (800 and 1300 °C) during 4 h.

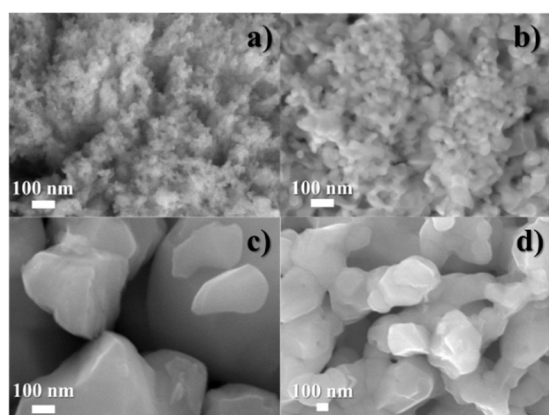
Figure 1a-c) shows FE-SEM images of SrSnO<sub>3</sub> spherical nanoparticles annealed at different temperatures. The average particle size is about 20, 65, and 500 nm for samples without annealing, annealed at 800, and 1300 °C, respectively. The SrSnO<sub>3</sub> sample annealed at 1300 °C present elongated shapes compared to the other two samples. Sm-doped samples did not show significant changes in the morphology, shape and size respect to the un-doped sample (Fig. 1d). The corresponding Sr, Sn and O maps present a homogenous distribution of these elements (Fig. 2). Structural properties were studied by Raman spectroscopy. Fig. 3 presents the Raman spectra of Sm (4%)-doped SrSnO<sub>3</sub> nanoparticles annealed at different temperatures. For sample without annealing, Raman peaks can be observed at 147 and 179 cm<sup>-1</sup> assigned to A<sub>g</sub> vibrational mode which is related to the Sn-O-Sn and O-Sn-O bonds, respectively. Other band located at 571 cm<sup>-1</sup> is associated to surface defects in SnO<sub>2</sub> nanocrystals, while the peaks at 701 and 1071 cm<sup>-1</sup> are related to presence of SrCO<sub>3</sub> [4]. At increased annealing temperature, intense Raman peaks at 114 (B<sub>2g</sub> mode), 223 (A<sub>g</sub> mode) and 257 cm<sup>-1</sup> (A<sub>g</sub> mode) are observed, which have been associated with orthorhombic phase of SrSnO<sub>3</sub> [5]. On the other hand, the peaks related to SrCO<sub>3</sub> decrease due to the annealing temperature. The PL properties of the samples were measured at room temperature, under excitation of a He-Cd laser (λ = 325 nm). Figure 4 shows PL spectra of Sm-doped SrSnO<sub>3</sub> annealed at 1300 °C, the spectra reveal a gradual increase in the intensity of transition levels of Sm<sup>3+</sup> (<sup>4</sup>G<sub>5/2</sub>→<sup>6</sup>H<sub>5/2</sub>, 560-590 nm, <sup>4</sup>G<sub>5/2</sub>→<sup>6</sup>H<sub>7/2</sub>, 600-630 nm) due to the rise of molar concentration [6].

This work highlights how the crystalline quality and phase stability are very important for luminescence properties enhancement in perovskites, which is often difficult to achieve. The Sm-doped SrSnO<sub>3</sub>

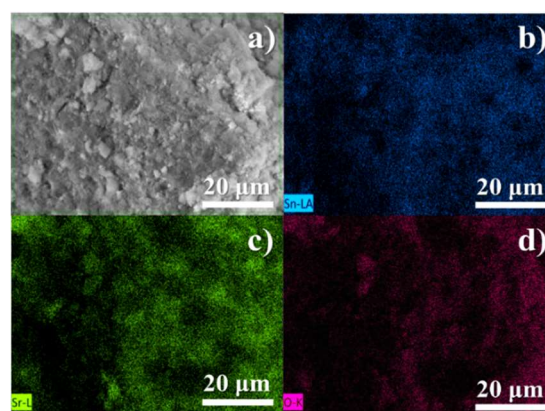
nanoparticles studied here exhibit intense red emissions, particle size less than 100 nm (after thermal annealing > 800 nm) and an orthorhombic stable phase [7].

#### References:

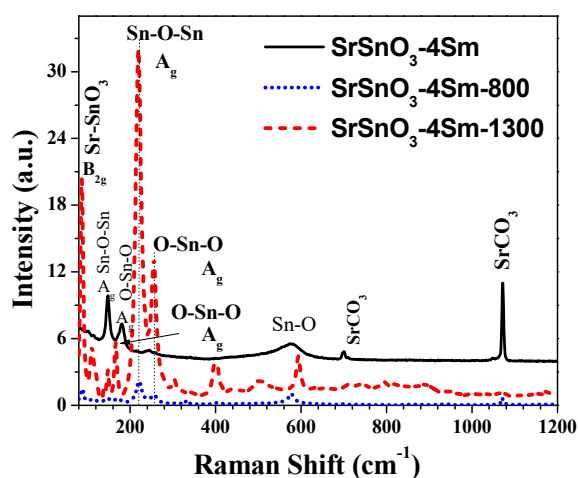
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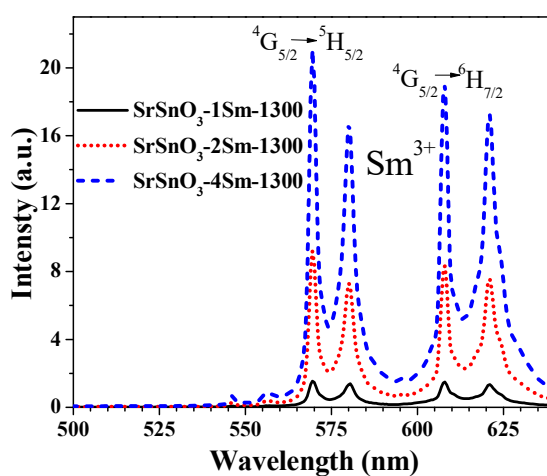
**Figure 1.** FESEM micrography of a)  $\text{SrSnO}_3$ , b)  $\text{SrSnO}_3$ -800, c)  $\text{SrSnO}_3$ -1300, and d)  $\text{SrSnO}_3$ -4Sm-1300.



**Figure 2.** Elemental mapping of a)  $\text{SrSnO}_3$  nanoparticles, b) Sn-LA c) Sr-L, and d) O-K.



**Figure 3.** Raman spectra of Sm-doped  $\text{SrSnO}_3$  nanoparticles annealed at different temperatures.



**Figure 4.** PL spectrum of Sm doped  $\text{SrSnO}_3$  nanoparticles annealed at 1300 °C.