ULTRA-SMALL CARBON SAMPLES AND THE DATING OF SEDIMENTS

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ABSTRACT. Radiocarbon dates from sediment core samples are usually obtained on the whole organic carbon portion in order to use as little of the valuable sediment as possible. Such measurements may not result in an accurate chronology of the sediment because the material may include carbon from different reservoirs at the time of deposition. The development of AMS techniques for dating ultra-small (20–500μg) carbon samples permits dating various components of cored sediments. We give examples of physical and chemical fractions of sediments which illustrate the large differences in ages obtainable from materials at the same sediment depth. The dating of all or several of these fractions demands more complex interpretations than just the derivation of a sedimentation rate from whole organic carbon dates.

INTRODUCTION

We have shown that ultra-small carbon samples (20–500μg) can be ^{14}C dated, but only with decreased total measurement precision due to uncertainties in processing contamination and background (Vogel, Nelson & Southon, 1987). However, the accuracy of the radiocarbon concentration measurement is not affected, and the date is often better associated with the event under investigation. This has been particularly beneficial in dating cored sediments where ^{14}C chronologies are usually obtained on "whole organic carbon." A precise date, using the least amount of valuable core material, is possible by measuring "whole organic carbon," but the result may be inaccurate. The dated whole carbon can often be a mixture from several reservoirs (fossil, benthic, terrestrial, etc) which have different ^{14}C ages. Further, the association of the date with the time of sedimentary deposition is not assured. We discuss here several cases in which good accuracy and association with sediment deposition were obtained by extracting and dating very specific physical or chemical carbon fractions of cored sediments.

MACROFOSSIL SELECTION AND DATING

Single-growth-year terrestrial macrofossils are physically separable carbon which are well suited to constructing accurate sediment chronologies. They have been in equilibrium with a known reservoir (the atmosphere) and generally have short transport times before deposition, depending on the watershed. Transport to the sediment was probably similar for identical macrofossils at different times. These materials also have the advantage of being immobile in the sediment, barring bioturbation, because of their size. As an example, we dated suites of chenopod seeds taken from a lacustrine sediment in western Canada (Table 1). A few

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seeds ($\approx 40\mu g$/seed) from this common family were sufficient for obtaining 1% (± 80yr) total measurement precision. Most lacustrine deposition phenomena take place on shorter time scales than 80 yr (Håkanson & Jansson, 1983). This precision, therefore, represented also the accuracy of the derived chronology.

TABLE 1

Examples of macrofossils used in dating sediments

RIDDL no.	Identification	Depth (cm)	Size (μg carbon)	Age (yr BP)	$\pm$
Lacustrine sediment					
−829	7 Chenopod seeds	372–382	250	1060	75
−830	8 Chenopod seeds	915–920	410	3600	75
Boreal climate indicators					
−815	2 *Betula papyrifera* seeds	825	52	10,430	380
−817	1 *Picea* needle	855	78	10,230	220
−819	5 *Betula papyrifera* seeds	875	48	10,470	440
−816	Unid. woody fragment	825	230	610	100
−818	Unid. woody fragment	855	120	3860	120
−819	Unid. woody fragment	875	160	3490	100
Charred barley grain					
−888	Barley seed		720	5220	100
−889	Barley seed acid wash		1000	5360	130
−890	Barley seed base wash		1200	5150	130
Fungus					
−867	Fungus cluster		215	20,600	1100
−868	Fungus cluster base wash		320	17,270	460

The identification of the macrofossils is helpful in establishing that materials are associated with the phenomena under study. Unidentified samples cannot be so judged for their contextual relevance. For example, a study of climatic fluctuations resulted in data in Table 1. The ages of unidentifiable woody materials disagreed with those for smaller, identified macrofossils which were found at the same levels. The macrofossils were direct indicators of the Boreal environment under investigation (Peteet *et al*, ms) whereas the woody materials were only thought to be associated with them. Mixing both materials from each level would have provided better measurement precision by increasing the sample size. Clearly, the results from such mixed samples would have been inaccurate with no association to the subject being studied. We now know that the woody fragments were introduced during coring. This was an extreme example of misplaced association threatening the accuracy of a chronology.

A more common problem is the contamination of macrofossils by chemically mobile contaminants. By comparing the ages of organic materials removed during acid and alkali washes with the age of the prepared sample, the level of contaminants remaining in the treated portion can be

estimated (Batten *et al*, 1986). Table 1 gives an example of a portion of a charred barley grain associated with an unlikely stratum in a sediment. The ages of the acid and base washes and the residue were within 1σ of each other, indicating that we accurately measured the age of the seed. Another example was a small fungus cluster found in clay sediments from the northern Queen Charlotte Islands of Canada (Table 1). The fungus was not as impermeable to contaminants as were seeds or cellulose. More than half the sample was soluble in base (0.25N NaOH) and this portion was 3ka younger than the insoluble part. However, the accurate age of the fungus would lie within the stated precision, even if up to 20% of the treated fungus were still composed of humic contaminants. Since the final base wash was not discolored, this was unlikely.
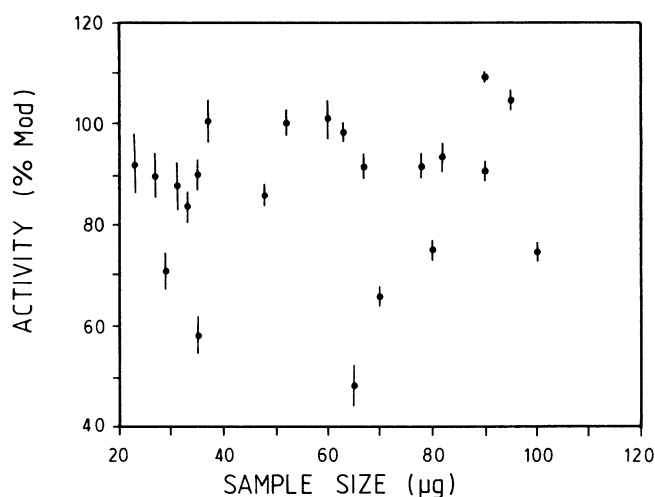

Fig 1. Final ^{14}C concentrations as a function of carbon sample size for 22 samples < 100 μg from a study of the ecosystem around a marine oil seep. [14]C concentrations were measured to a precision of 1–2%. Final precisions were 3–8% after size-dependent background subtraction.

We recently studied living meio- and macrofauna from cored sediments in order to follow fossil carbon through the ecosystem around an oil seep (Bauer, Spies & Vogel, ms). Dissolved inorganic carbon in 1-5cc of porewaters from these sediments were also analyzed, as were lipids and total organic carbon. Figure 1 shows the activity and final precision for the 22 samples containing 20–100μg of carbon in this study. Precisions of 1 to 2% were obtained in the ^{14}C concentration measurements, with total measurement precision of 3–8% arising from the uncertainty in the background subtraction. These precisions were entirely adequate to distinquish the path of fossil carbon through the local ecosystem. The use of ultra-small samples provided details which would have been masked by larger, less specific samples.

CHEMICAL FRACTIONS IN MARINE SEDIMENTS

Two cores, one from the Black Sea and one from the Gulf of California, were used to study chemical fractions in marine sediments.

Black Sea Core

The Black Sea core (Atlantis II, #1443, Ross & Degens, 1974) was previously used in a study of sapropel formation and only a few hundred milligrams of sediment from each level were available for analysis. We studied a base soluble fraction (0.25N NaOH with sonication) and the residual sediment along with the original whole organic carbon dates (Table 1 and Fig 2). The depth of the carbon-rich sapropel is obvious from the carbon con-
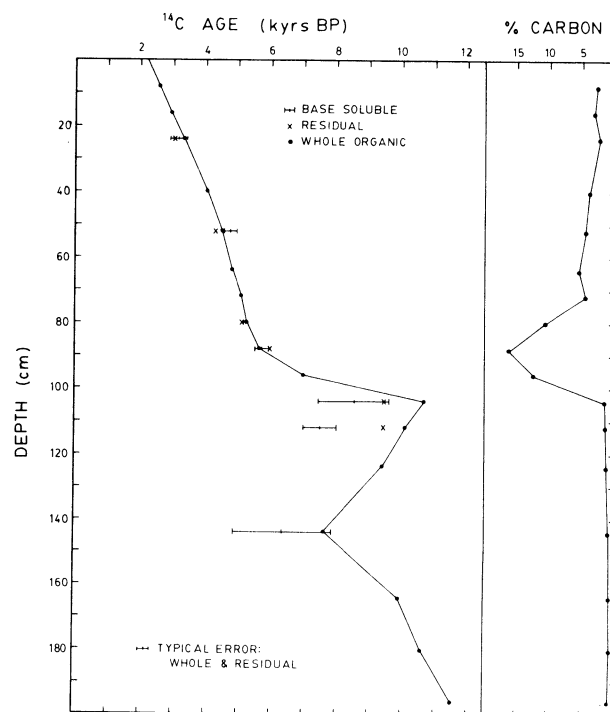

Fig. 2. The ^{14}C age *vs* depth for whole organic carbon, base-soluble fraction, and residual sediment from core 1443 from the Black Sea. The organic carbon content of the core is also shown, indicating the carbon-rich, sapropelic layer. Humic material from the sapropel diffused into the older carbon sediment below 100 cm.

centration. We have seen older carbon in the sediment below a sapropel in another Black Sea core (Calvert, Vogel & Southon, 1987) but not as distinctly as in this example. The transition from the older layer to the sapropel provided a step function around which carbon mobility could be studied, since the anoxic conditions of the Black Sea prevent benthic mixing. Humic material was separated from the whole sediment for 7 of the 17 available

strata. Sample sizes for these humic samples ranged from 260 μg down to 18μg, and the final measurement precisions reflected these sizes. The humic fractions in and above the sapropel were equal in age to the whole organic carbon, indicating an autochthonous source. The humics extracted from the older carbon-poor layer below the sapropel were significantly younger than the whole organics. This suggested that the base soluble carbon from the overlying sapropel was mobile and diffusing into the older sediment, as

TABLE 2

Black Sea Core 1443 organic carbon ages

RIDDL no.	Identification	Depth (cm)	Size* (μg carbon)	Age (yr BP)	±
−285	Whole organic	24		3310	200
−1043	Base soluble fraction		65	3140	240
−1044	Residual sediment			3060	90
−287	Whole organic	52		4490	170
−375	Base soluble fraction		260	4720	200
−1045	Residual sediment			4290	140
−290	Whole organic	80		5240	90
−1046	Base soluble fraction			5210	80
−1047	Residual sediment			5140	100
−291	Whole organic	88		5660	170
−377	Base soluble fraction		670	5680	170
293	Whole organic	104		11,130	100
−1048	Base soluble fraction		22	8530	1080
−1049	Residual sediment			9460	120
−294	Whole organic	112		10,080	210
−379	Base soluble fraction		50	7490	500
−1050	Residual sediment			9000	120
−296	Whole organic	144		7620	170
−1051	Base soluble fraction		18	6340	1500
−1052	Residual sediment			7670	120

*Size is given only for samples < 1000μg.

could be expected in the basic (pH $\approx$ 8.1) marine environment. The data were compatible with a diffusion length of $\approx$ 10cm, assuming the humic content of the sediments to be 10%.

The "residue" sediment after humic extraction was generally younger than the whole organic carbon date, but significantly so only for the two strata below the sapropel. This may have been a contamination artifact of the base wash. The extraction was not done in an inert atmosphere, although all portions were well acidified prior to combustion. Atmospheric exchange with organic carboxylic acids during the base extraction was tested

on a 130ka sediment from the Panama basin. The non-humic residue was 4.5% more modern than the inorganic fraction, more than enough to explain the smaller differences in the Black Sea core.

Gulf of California Core

The core from the Gulf of California (DSDP or ODP#480) was supplied in 3cc portions. The samples were thin sections taken across the core at well-defined varves, as indicated by nuclear magnetic resonance imaging (Briskin *et al*, 1986). An aliquot of 1–1.5g from each depth was used to obtain various organic fractions. The lipids were removed by dissolution in chloroform during 3 20-min rinses with sonication. The chloroform was slowly evaporated from both the extracted lipids and the residual sediment in air and further evaporated under vacuum at 80°C overnight. The sediment was then hydrolyzed for 1 hr with sonication in 6N HCl. The hyd-
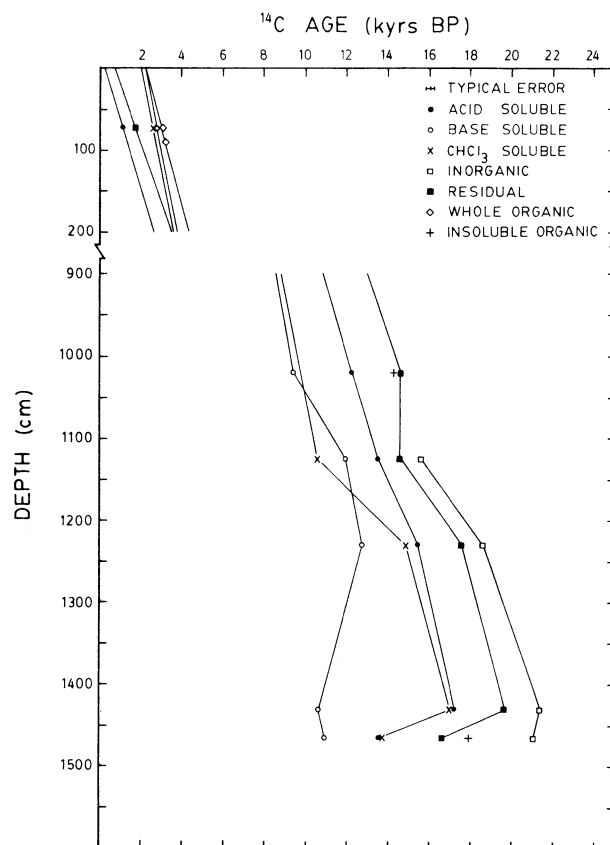

Fig. 3. The ^{14}C age *vs* depth for chemical fractions of the carbon in ODP core 480 from the Gulf of California. The wide disparity in ^{14}C ages at each depth demonstrated the chronologic inaccuracy introduced by dating the whole organic carbon from such a core.

rolysate was brought to pH ≈ 5 using NaOH, centrifuged, decanted slightly and evaporated to dryness. The rinsed sediment was then sonicated in 0.1N NaOH plus 0.1M $Na_4P_2O_7$ for 1hr to extract humic materials. Ramunni and Palmieri (1985) showed that such treatment is equivalent to a 24-hr soak. The solute was acidified and the precipitate was separated and dried. The final sediment residue was also acidified and dried. All materials were combusted in sealed tubes with CuO at 900°C. The bulk inorganic carbonate was dated by hydrolyzing a portion with 85% H_3PO_4.

In contrast to the Black Sea sediments, components of the Gulf of California core possessed an extremely wide range of ages at each depth, as shown in Figure 3. No unique age could be assigned to the strata, but ^{14}C could be used as a geochemical tracer to understand the processes at work in the Gulf. The studied interval was chosen to include the Pleistocene/ Holocene boundary as indicated by preliminary dating (Shackleton Hall, 1982; Spiker & Simoneit, 1982). The influx of older material at ca 1450cm was probably due to increased erosional activity at that time. The similar trends in the ages of the individual chemical fractions indicated a relatively stable but allochthonous source of carbon in this sediment through the early Holocene.

Whole organics were dated for two shallow strata and were older than the extracts. At the shallowest stratum, the residual was younger than the whole organic carbon and two of the extracted components, although no balancing old component was found. Exact sonochemical effects on sediment extractions have not been well researched and may be responsible. The largest fraction of carbon was the acid/base/solvent insoluble residue. This residue was the oldest organic component at early Holocene depths and was separable gravimetrically into components attached and unattached to clay particles. Even these two fractions had detectably different ages ("Residual" and "Insoluble Organic" in Fig 3). Carbon compounds which were that immune to separation from clay particles could have been on these particles long before deposition. If such compounds are dominant in a "whole organic carbon" date, they provide only an upper limit to the time of deposition.

Except near the top of the core, humic materials were younger than other fractions which are their presumed chemical precursors (Ishiwatari, 1985; Vandenbroucke, Pelet & Debyser, 1985). These data and the high ^{14}C activity of that fraction at 1430cm could be explained by the mixing of materials of different ages. Exchange with atmospheric CO_2 during base extraction was insufficient to explain the extreme youth of this material. As with the Black Sea sediment, the humic portion appeared to be mobile. Our data did not cover sufficient distances to determine a diffusion length, which must be large, judging by the youth of the base soluble fraction and the high sedimentation rate (≈ 100cm/kyr). The mobility of this fraction indicated that humic acids must either be dated separately or removed before dating a marine sediment core.

The components (amino acids, carbohydrates, etc) released by acid hydrolysis (Degens, Reuter & Shaw, 1964) were younger than the clay-attached carbon but older than the base or solvent soluble fractions. This fraction is chemically imprecise and dependent on the hydrolysis procedure (Vandenbroucke, Pelet & Debyser, 1985). The hydrolysates were the

smallest fractions extracted, 120 to 220μg of carbon, but this fraction had the most stable age/depth trend and the lowest extrapolated age at the sediment surface. More research is needed to determine if this fraction could yield accurate chronologies.

Fowler, Gillespie and Hedges (1986) suggest that lipids are the best material for dating sediments. Lipids are primary compounds produced by plants and animals, and are in near equilibrium with atmospheric or marine CO_2 at the time of production. They are less mobile than inorganically soluble molecules. Ishiwatari (1985) shows that lipids are a precursor to humic and humin substances. Except for reservoir effects, they should be the youngest immobile fraction in a sediment, as they were here. They should provide the most accurate measure of the time of deposition. Since 50-100μg of carbon extracted as lipids from a single core can be accurately dated, "bulking samples from several cores" (Fowler, Gillespie & Hedges, 1986, p 449) to obtain large amounts of lipids is not necessary. The larger sample may give better measurement precision, but the bulking process might mix lipids of different ages, destroying the accuracy of the result.

The ODP 480 core showed that several carbon fractions with disparate ^{14}C concentrations can be separated from some sediments. Only one, or perhaps none, of these fractions may provide an accurate chronology of deposition, while the others provide evidence of their source or transport. In any event, a ^{14}C age taken on whole organic carbon from such a sediment will be inaccurate, no matter how precise the ^{14}C measurement.

CONCLUSION

Core chemistries and macrofossils must be studied carefully so that a carbon fraction derived from a known reservoir may be selected for dating purposes. The more identified a fraction is, the smaller it will be. Ultrasmall sample techniques are now making this selectivity possible, resulting in significant improvements in the accuracy and the association of ^{14}C measurements on sediment cores.

ACKNOWLEDGMENTS

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