

TEM Characterization of W-O-N Coatings

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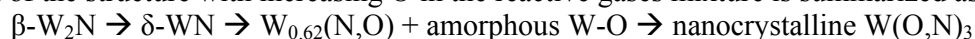
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Recently, a new class of coatings based on oxynitrides has drawn much attention in the research field as well as in industrial applications, as shown by either the large numbers of recent publications on TM-O-N systems (TM - transition metal) such as Ti-O-N, Zr-O-N and Ta-O-N, or the development of Si-O-N for opto-electronic devices [1]. The properties of these coatings are related to the chemical composition and the structural arrangement. However, the knowledge about the structure of TM-O-N systems is very limited, especially how the structural arrangement of the non-metallic elements is in the lattices. To the best of our knowledge, only a few studies exist on the development of structural models for oxynitrides, based on XRD and/or XPS analysis, as e.g. Si-O-N [2] and Ti-O-N [3], or on Mössbauer spectrometry for Fe-O-N [4]. TEM was used scarcely for the characterization of TM-O-N coatings possibly due to the damage of the structure by the electron-irradiation as it is reported for Cr-O-N [5]. This work is aimed at the crystallographic understanding of W-O-N sputtered films by using TEM and HR-TEM techniques for complementing the information provided by XRD characterization.

W-O-N coatings were prepared by DC reactive magnetron sputtering from a pure tungsten target in an Ar + O₂ + N₂ atmosphere. Starting from the Ar + N₂ atmosphere, the progressively increased flow of oxygen to the discharge allowed varying the chemical composition of the coatings from high to low N/O ratios. The chemical composition and the structure of the films were analysed by electron probe microanalysis (EPMA) and X-ray diffraction (XRD - with Co K α radiation), respectively. The structural analysis was complemented by (high-resolution) transmission electron microscopy and electron diffraction (HRTEM / ED - operating at 200 kV). Further details on the deposition procedures can be found in previous authors papers [6, 7]

Fig. 1 shows the chemical composition and XRD patterns of W-O-N coatings. As expected, there is an inverse trend between W and O contents. That is to say, the content of W steadily decreases whereas the O content increases with increasing O₂ flow rate, until reaching the composition (24 at.% W and 71 at.% O) very close to the stoichiometry of WO₃ compound. However, for the same O₂ flow rate variation, the N content only starts to decrease at $fO_2 = 6$ sccm, below of which the N content remains approximately constant. These results shows that in all the coatings the ratio between W and N/O is lower than the one regarding the stoichiometry of WN and WO₃ compounds [6]. With increasing fO_2 a progressive decrease of the crystallinity could be detected with some coatings even showing typical XRD-amorphous patterns. For fO_2 up to 6 sccm (N \geq 47 at.%) several well-defined diffraction peaks could be detected at 2θ close to 42, 50 and 75°. These peaks are indexed to the nitrogen rich phases, β -W₂N, δ -WN or W_{0.62}(O,N), as it was discussed in Ref. [6]. Moreover, for $fO_2 \geq 3$ sccm, the coatings showed a broad diffraction peak centred at $2\theta \approx 30^\circ$ which is attributed to a quasi-crystalline W-O phase. For W-O-N coatings deposited with $fO_2 \geq 7$ sccm, the structure exhibited a nanocrystalline WO₃ structure [6]. The evolution of the structure with increasing O in the reactive gases mixture is summarized as:



From the XRD results, two main points remain unclear and have motivated further characterization of selected coatings by more sophisticated techniques:

- 1) What is the volume fraction of the crystalline nitride phase and the amorphous W-O phase in the $W_{27}O_{26}N_{47}$ coating?
- 2) What is the crystallographic difference between the coatings $W_{24}O_{38}N_{38}$ and $W_{24}O_{72}N_{04}$ that exhibited similar XRD patterns but different N contents?

For the coating $W_{27}O_{26}N_{47}$, the electron diffraction pattern shown in Fig. 1b confirms the XRD analysis and reveals the presence of (oxy)nitride nanocrystallites. The sharp ring patterns at the reciprocal distance of 4.18, 4.85, 6.86, 8.05 and 9.70 nm^{-1} fit perfectly to the first five strong diffractions of both cubic β - W_2N phase (ICDD card No. 25-1257) and $W_{0.62}(N,O)$ phases (ICDD card No. 25-0254 & 89-4762). Because of the high oxygen content, $W_{0.62}(N,O)$ phase seems to be more favoured in this case. In addition, there are several rather broad/diffuse rings at the reciprocal distance of 3.15, 9.7, 10.9 and 12.0 nm^{-1} , of which the first one corresponds to an interplanar distance of 0.317 nm. This distance is two times of the W-O bonds when one oxygen atom is singly bonded to the tungsten atom, or corresponding to the strongest (200) diffraction of hexagonal WO_3 phase (ICDD card No. 85-2459). The former case can only occur in amorphous WO_3 where oxygen is deficient or in the presence of impurities such Na^+ [8]. The large ratio between O+N and W (>2.5) suggests that N is present in the amorphous WO_3 phase and thus hinders the crystallization. The HR-TEM micrograph in Fig. 2 reveals that nanocrystallites of about 8 nm in diameter are embedded in the amorphous matrix. With numerical analysis, the FFT images from the crystalline and quasi-amorphous regions are shown in the inserts (a) and (b) of Fig. 2. From the insert (a), it is possible to recognize an f.c.c. NaCl structure by the {111}, {200} and {220} plans. However the small difference in the lattice parameter between the β - W_2N and the $W_{0.62}(N,O)$ phases (0.4126 nm versus 0.4138 nm) does not allow an unambiguous identification.

In the case of coating $W_{25}O_{38}N_{37}$, the analysis by XRD and electron diffraction gives the same results: no long range ordering can be identified. By deconvolution of the intensity line profile it is possible to distinguish three diffuse rings centred at 2.66, 3.8 and 5.6 nm^{-1} , corresponding to interplanar distances of 0.375, 0.263 and 0.179 nm (Fig. 1c), which can be identified as the plans {200}, {220} and {400}, respectively, of m- WO_3 phase (ICDD card No. 83-0950). Moreover, a more intense ring centred at 3 nm^{-1} represents the amorphous WO_3 phase, as discussed in the previous paragraph. Even using HT-TEM (see Fig. 3a), only some "clusters" with a diameter smaller than 2 nm, that are difficult to identify, suggest the existence of short range order (the existence of the diffuse diffraction rings). The no-detection of any ring (even diffuse) that could be indexed as a W-N phase suggests that N is saturated in the W-O nanocrystalline/amorphous phase.

The coating $W_{24}O_{71}N_{05}$ with similar XRD pattern but lower nitrogen content was also analysed by HR-TEM. Exactly the same aspect as $W_{25}O_{38}N_{37}$ was achieved, when high resolution and FFT were used. However, after a few minutes of e-beam irradiation, a structural modification was observed. Such a phenomenon has already been observed in W-O [9] or in Cr-O-N [5] thin films. However, it should be pointed out that this modification did never happen in the TEM study of other W-O-N coatings, particularly in the coating $W_{25}O_{38}N_{37}$. After the crystallization (Fig.3b), the structure of the coating $W_{24}O_{71}N_{05}$ can be identified as the orthorhombic W_3O_8 phase (ICDD card No. 81-2262), where the reflections of the plans {130} and {200} were indexed. This different behaviour suggests that the presence of higher amounts of N in the quasi-amorphous WO_3 phase, as in the coating $W_{25}O_{38}N_{37}$, contributes to its stability and retards crystallization.

In this work, the structural evolution of W-O-N coatings could be summarized as: β - $W_2N \rightarrow \delta$ -WN $\rightarrow W_{0.62}(N,O) +$ amorphous W-O \rightarrow nanocrystalline $W(O,N)_3$. Further studies using HR-

TEM showed that: (1) $W_{27}O_{26}N_{47}$ is composed of nanocrystalline $W_{0.62}(N,O)$ grains embedded in a quasi-amorphous oxide matrix, (2) $W_{25}O_{38}N_{37}$ is characterized by a full amorphous structure and, (3) $W_{24}O_{71}N_{05}$ exhibits also an amorphous structure but may crystallize after some minutes of e-beam irradiation, suggesting that the saturation of nitrogen could stabilize the amorphous phase.

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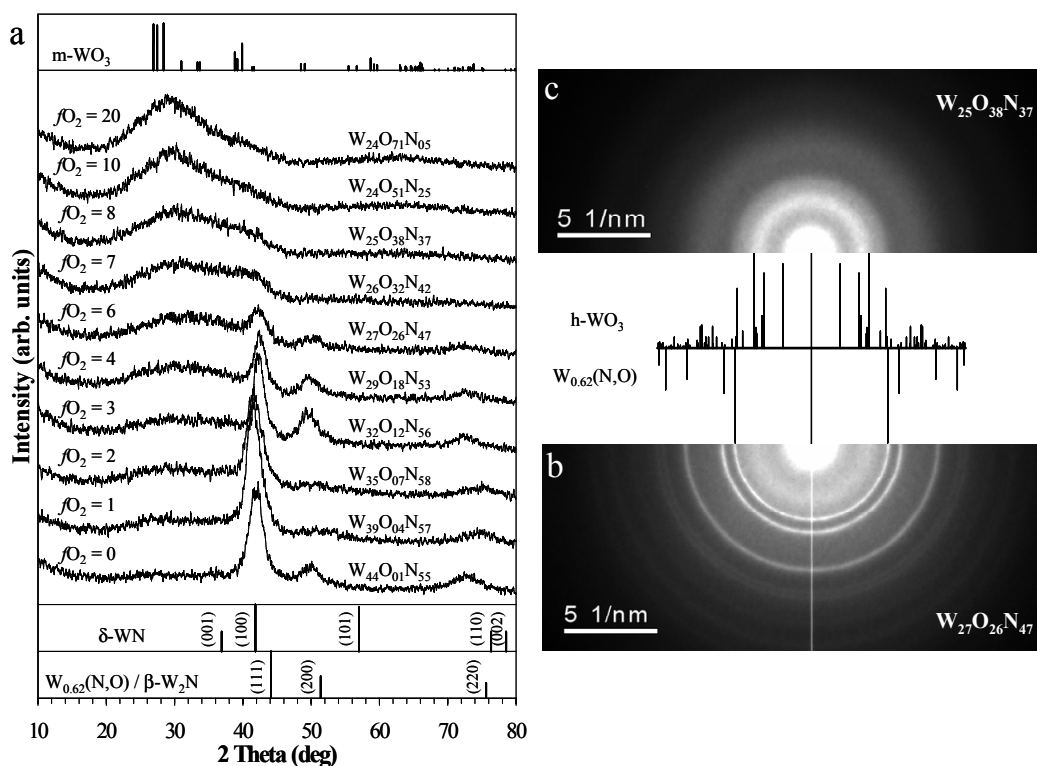


Fig. 1. (a) XRD pattern of the W-O-N coatings deposited with increasing oxygen flow [6]. Selected area electron diffraction patterns of the coatings $W_{27}O_{26}N_{47}$ (b) and $W_{25}O_{38}N_{37}$ (c).

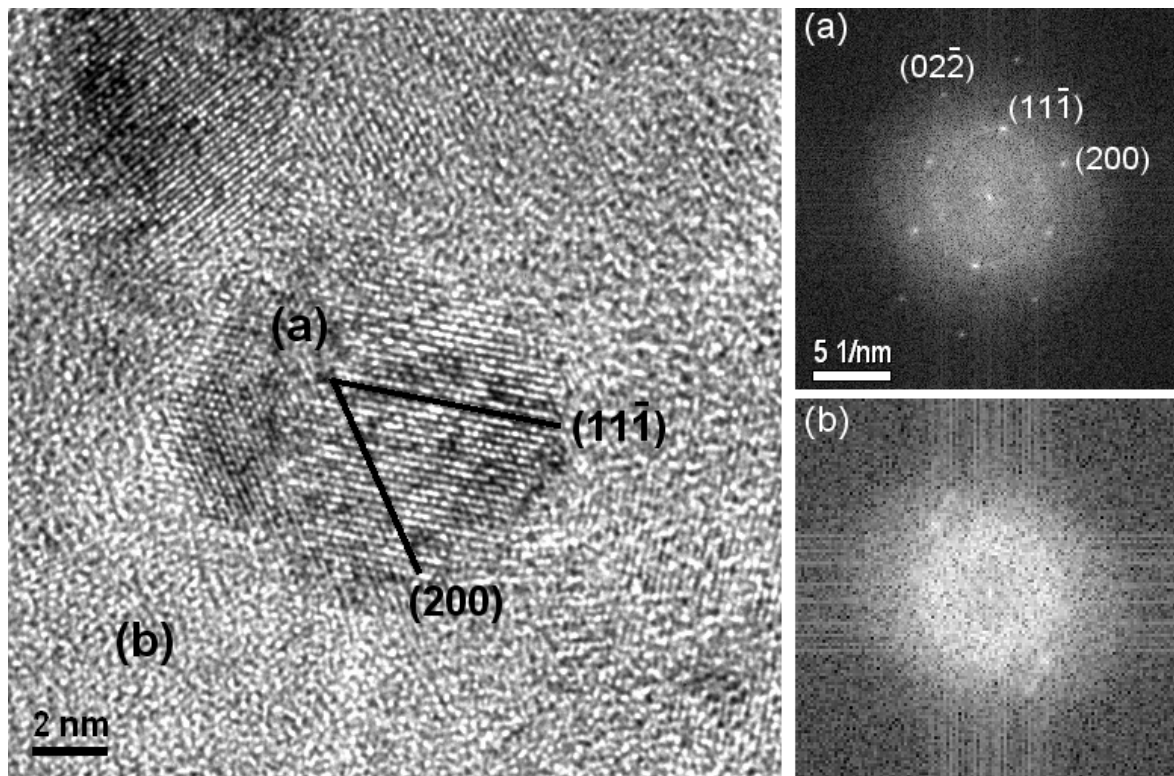


Fig. 2. HR-TEM micrograph of coating $W_{27}O_{26}N_{47}$ showing nanocrystallites embedded in amorphous matrix. The inserts (a) and (b) show the diffraction patterns obtained by FFT on the crystalline and quasi-amorphous areas indicated in the micrograph.

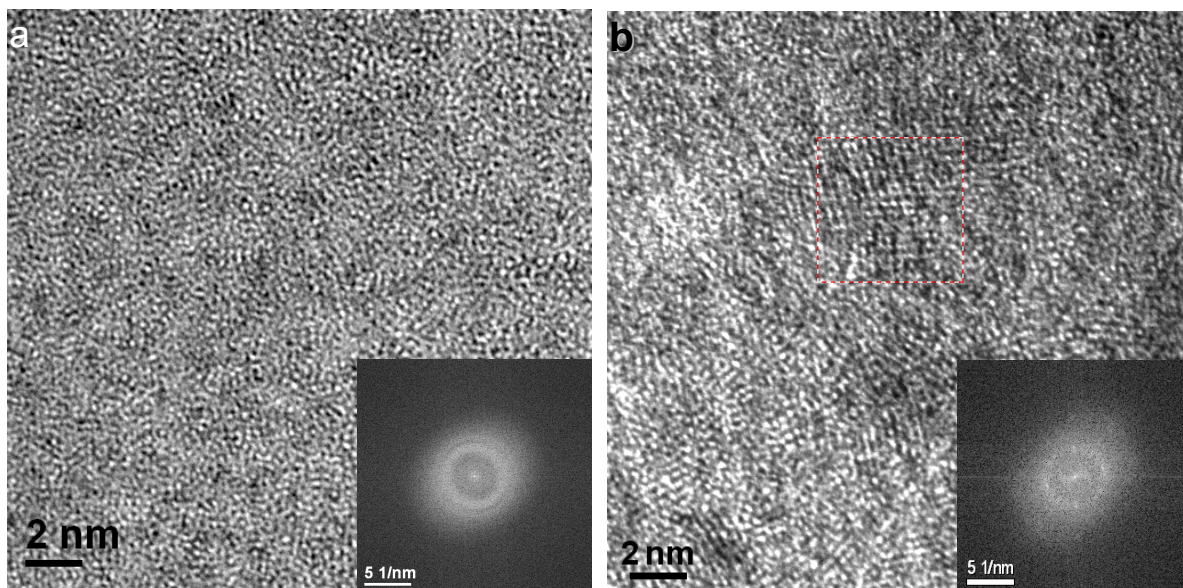


Fig. 3. HR-TEM micrograph of coatings: (a) $W_{25}O_{38}N_{37}$ and (b) $W_{24}O_{71}N_{05}$ partly crystallized due to the irradiation of electron beam.