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¹⁴C IN TREE RINGS IN THE VICINITY OF THE RBMK REACTOR NUCLEAR POWER PLANT

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ABSTRACT. The paper presents the results of radiocarbon (14 C) concentration measurements in tree rings in the vicinity of Kursk NPP (Russia) with four operating RBMK reactors. The sampling was carried out from the site with the highest expected accumulation of radiocarbon in vegetation. The site was determined with long-term meteorological data. The measurements of 14 C concentration carried out with accelerator-mass spectrometer in Budker Institute of Nuclear Physics, Novosibirsk, Russia. The obtained results demonstrated the influence of exploitation of Kursk NPP to the concentration of 14 C in tree rings. Based on the equilibrium between the 14 C ratio in the tree rings and the surrounding air, retrospective estimates of the radiocarbon discharge and effective doses were made. Effective doses were calculated with two approaches: IAEA methodology and less conservative approach, considering the real food consumption in the Kursk region. The values of calculated doses by the second method (0.08–2.58 µSv) are more than 2 times less than IAEA approach (0.17–5.30 µSv). The highest difference between measured and background 14 C in tree ring is 41.7 ± 5.8 pMC in 2014 during the restoration of graphite stack. The main contribution to 14 C exposure in the considering period is caused by background – from 70 to 99%.

KEYWORDS: carbon-14, effective dose, radioactive discharges, RBMK, retrospective assessment.

INTRODUCTION

Carbon-14 (¹⁴C) is a low-energy beta-emitting radionuclide that is produced both naturally and artificially. The natural path is based on the interaction of thermal neutrons and nitrogen-14 atoms in the upper layers of the troposphere and stratosphere: ¹⁴N(n,p)¹⁴C. About 1.4·10⁶ GBq ¹⁴C is produces annually by this reaction, while the total amount of ¹⁴C in the atmosphere is estimated at 1.4·10⁸ GBq (IAEA 2004). Nuclear weapons testing led to an almost twofold increase in the concentration of ¹⁴C in the atmosphere. According to estimates, 2.2·10⁸ GBq ¹⁴C entered the atmosphere in the 1950s and 1960s (IAEA 2004). After acceptance of Partial Nuclear Test Ban Treaty ¹⁴C concentration in atmosphere began to decline gradually due to the carbon cycle and is now comparable to pre-test levels ~ 100 pMC ≈ 226 Bq/kg C (IAEA 2004).

At present, atmosphere discharges from nuclear fuel cycle enterprises are the main anthropogenic source of ¹⁴C entering the atmosphere. During the operation of a nuclear reactor, ¹⁴C is formed mainly as a result of the interaction of thermal neutrons and ¹⁴N, ¹³C, ¹⁷O atoms present in fuel elements, structural materials, moderator and coolant, as well as due to uranium and plutonium ternary fission reactions in nuclear fuel (IAEA 2004). It is estimated that about $1.1 \cdot 10^5$ GBq/year is released into the atmosphere in gaseous form from all operating nuclear power plants, while about $3.7 \cdot 10^5$ GBq/year of ¹⁴C in both gaseous and liquid form released by spent nuclear fuel reprocessing plants.

In Russia, there are three types of nuclear power reactors: VVER (PWR), FBR and RBMK. In an RBMK ¹⁴C is mainly produced in the graphite moderator by ¹³C(n,γ)¹⁴C and ¹⁴N(n,p)¹⁴C reactions, due to large amounts of impurities. Most of the produced ¹⁴C, however, remains in the graphite during the operation of the reactor and is not released to the atmosphere.



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Figure 1 Wind rose in Kurchatov 2006–2022.

Tree rings are a good indicator of the ¹⁴C content in the atmosphere due to the process of photosynthesis. The analysis of ¹⁴C content in the annual rings of trees located in the vicinity of nuclear facilities allows to perform a retrospective assessment of ¹⁴C release and calculate effective doses throughout the entire life cycle of the enterprise. An increased content of ¹⁴C was observed after the analysis of tree rings near Paks (Janovics et al. 2013), Ignalina (Ežerinskis et al. 2018), Barseback (Stenstrom et al. 1997), and Beloyarsk (Nazarov et al. 2022) NPPs. The aim of this paper is retrospective annual discharges of ¹⁴C by Kursk NPP and effective doses calculation.

SAMPLING AND MEASUREMENTS

Kursk NPP is presented by four RBMK-1000 reactors with an electrical output of 1000 MW. Four reactors were constructed from 1979 to 1986. One of four reactors is permanent shutdown since 2021. The routine release of ¹⁴C into the atmosphere is carried out mainly through two ventilation stacks 150 m high and from other organized sources, but with lower concentrations. The sampling site, located 2.7 km west of the geometric center of discharges sources, was defined using long-term meteorological data (Rostekhnadzor 2021). The weather archive since 2006 included more than 50,000 records. The dominant wind direction is east-southeast (Figure 1). Figure 2 presents the location of Kursk NPP and the sampling site.

The core samples of *Pinus sylvestris* L. and *Populus tremula* L., the most representative tree species in the location of Kursk NPP, were sampled using an increment borer at a height of about 130 cm from the soil surface. The age of the trees varied from 50 to 70 years. The annual ring widths on all cores were measured according to standard dendrochronological techniques (Stokes and Smiley 1968; Cook and Kairiukstis 1990; Rinn 1996). We used COFECHA software (Holmes 1983) cross-dating quality control. According to the crossdated series cores were divided into annual rings. Rings of the same age from 10 different individuals were combined into one sample corresponding to a certain year. 14 different samples were selected for analysis. The ¹⁴C concentration of a 113-year-old pine tree located in Novosibirsk was used as background (Nazarov et al. 2021) due to the city of Novosibirsk is located at a considerable distance from the operating nuclear facilities.



Figure 2 The sampling site and Kursk NPP location (numbers of power units).

During the sample preparation, cellulose was extracted from tree rings by the procedure described in details in (Parkhomchuk et al. submitted). In general the procedure consisted in several steps: (1) multistage extraction purification with $C_2H_2Cl_2$, C_2H_5OH and distilled water until a colorless solution was obtained; (2) delignification, which was carried out by the catalytic oxidation by H_2O_2 in an acidic medium with NaWO₄ as a catalyst; (3) multistage dehumification with NaOH; and finally (4) acid treatment with HCl for CO₂ removal, distilled water rinsing and drying at 70°C with snow-white cellulose obtaining. Then it was subjected to complete combustion and transformation into graphite-like carbon in the absorption-catalytic setup (Lysikov et al. 2018), with following measurements of ¹⁴C concentration. The measurements were carried out on accelerator mass-spectrometer at the AMS Golden Valley laboratory using the facility from Budker Institute of Nuclear Physics (Parkhomchuk and Rastigeev 2011; Nazarov et al. 2021; Petrozhitskiy et al. forthcoming).

METHODS

The used calculation approach assumes that the concentration of ${}^{14}C$ in the annual ring of a tree corresponds to the average annual concentration in the ambient air. The sampling site is expected to have the highest concentration of ${}^{14}C$ in the air and, as a result, the highest effective dose.

The ¹⁴C air concentration A_V (Bq/m³) was calculated using following formula:

$$A_V = \Delta p M C \cdot S \cdot C_{st},\tag{1}$$

where ΔpMC – difference between measured and background ¹⁴C concentration (Table 1);

S – coefficient between pMC and Bq/kg C. 100 pMC = 226 Bq/kg C;

 C_{st} – stable carbon concentration in air. The concentration of stable carbon was determined using carbon dioxide levels in atmosphere (NASA 2023). The carbon dioxide concentration

1346 E I Nazarov et al.

| Year | ¹⁴ C concentration (pMC) | | Stable carbon in air concentration | |
|------|-------------------------------------|-----------------|------------------------------------|--|
| | Kursk NPP | Background | (g C/m ³) | |
| 1976 | 145.7 ± 1.2 | 134.8 ± 1.3 | 0.173 | |
| 1980 | 134.8 ± 0.7 | 126.5 ± 3.2 | 0.177 | |
| 1984 | 128.7 ± 0.7 | 125.3 ± 2.8 | 0.180 | |
| 1986 | 125.7 ± 0.8 | 121.9 ± 3.3 | 0.182 | |
| 1988 | 124.0 ± 0.7 | 115.2 ± 1.8 | 0.184 | |
| 1990 | 121.8 ± 0.7 | 114.8 ± 2.3 | 0.185 | |
| 1994 | 116.8 ± 1.1 | 112.0 ± 2.2 | 0.187 | |
| 1998 | 112.7 ± 0.7 | 109.9 ± 1.5 | 0.192 | |
| 2002 | 112.4 ± 0.8 | 105.7 ± 2.1 | 0.195 | |
| 2006 | 117.8 ± 0.7 | 103.2 ± 1.2 | 0.200 | |
| 2010 | 107.7 ± 0.7 | 102.5 ± 0.9 | 0.204 | |
| 2014 | 143.5 ± 5.8 | 101.8 ± 0.3 | 0.208 | |
| 2018 | 136.7 ± 2.7 | 100.7 ± 0.4 | 0.213 | |
| 2020 | 101.3 ± 0.7 | 100.0 ± 0.4 | 0.216 | |

Table 1 Concentrations of ¹⁴C in tree rings and ¹²C in air.

Table 2Dilution factor values for each year.

| Year | 1976–2002 | 2006 | 2010 | 2014 | 2018 | 2020 |
|-----------------------------------|-----------|------|------|------|------|------|
| $G_{n,j,} \ 10^{-8} \ { m s/m^3}$ | 4.74 | 5.44 | 6.06 | 7.45 | 5.65 | 4.95 |

varied from 332 ppm in 1976 to 414 ppm in 2020. The values of C_{st} used in calculations are presented in Table 1.

The estimate of the annual ¹⁴C discharge Q (Bq/a) was calculated by formula:

$$Q = 3.15 \cdot 10^7 A_V / G_{n,j},\tag{2}$$

where $3.15 \cdot 10^7$ – the conversion factor, s/a;

 $G_{n,j}$ – the annual average meteorological dilution factor in the surface layer of the atmosphere at a distance x from the *i*-th source in the wind direction of the *n*-th point, s/m³ (Figure 1). For sampling site, the values were calculated for each year, where it possible, by analyzing the meteorological data (Table 2).

The IAEA approach for estimating annual effective doses from ¹⁴C releases assumes that all the food consists only of local products and, as a result, there is an equilibrium in the ¹⁴C concentration in food, human tissue, and atmospheric air. Then the estimate of the expected annual dose of internal exposure D (Sv/a) from ¹⁴C contained in the atmospheric air in the form of carbon dioxide ¹⁴CO₂ can be obtained by the following formula (IAEA 2001):

$$D = A_V G / C_{st},\tag{3}$$

| Product | $R_i (\mathrm{kg})^{\mathrm{a}}$ | $\alpha_i (\%)^a$ | $f_{p,i} (\text{kg C/kg})^{\text{b}}$ |
|------------|-----------------------------------|-------------------|--|
| Milk | 291.7 | 34.3 | 0.065 |
| Meat | 93.7 | 45.7 | 0.2 |
| Bread | 89.5 | 50.0 | 0.39 |
| Potato | 103.4 | 90.7 | 0.046 |
| Vegetables | 101.9 | 89.3 | 0.059 |
| Fruit | 81.2 | 93.6 | 0.062 |

 Table 3 Parameters of products consumption in Kursk region.

^aRostekhnadzor (2021).

^bIAEA (2012).

where G – is the effective dose rate factor that relates the annual dose rate (Sv/a) to the concentration of ¹⁴C per gram of carbon in people (Bq/g). The dose rate factor recommended for screening is $5.6 \cdot 10^{-5}$ Sv/a per Bq/g (IAEA 2001).

Since ingestion is the primary mode of expose, and the share of consumption of local products does not always exceed the share of imported, it is reasonable to assume that there may not be an equilibrium of ¹⁴C. In this case, the expected annual dose of ¹⁴C exposure can be calculated using the formula (Kryshev et al. 2020):

$$D = \varepsilon_{ing} U A_V + \varepsilon_{food} \sum_i \alpha_i R_i A_{food,i}, \tag{4}$$

where $\varepsilon_{ing} = 6.5 \cdot 10^{-12}$ Sv/Bq – dose conversion factor for an adult when inhaling ¹⁴C in the form of CO₂ (ICRP 2012);

 $U = 8.1 \cdot 10^3$ m³/a – respiratory rate of an adult (Pleil et al. 2021);

 $\varepsilon_{food} = 5.8 \cdot 10^{-10}$ Sv/Bq – dose conversion factor for dietary intake of ¹⁴C for an adult (IAEA 2014);

 α_i – the share of local products in the diet of the population (Table 3);

 R_i – annual consumption of the product by the population, kg (Table 3);

 $A_{food,i}$ – ¹⁴C concentration in the local food product, Bq/kg.

The ¹⁴C concentration in the food product, based on the balance of ¹⁴C between atmospheric air and local plant or animal products, was calculated by the formula:

$$A_{food,i} = A_V f_{p,i} / C_{st},\tag{5}$$

where $f_{p,i}$ – the share of stable carbon in plant or animal product, kg C/kg product (Table 3).

RESULTS AND DISCUSSIONS

The ¹⁴C concentration in tree rings near Kursk NPP are presented in Figure 3. The launch of four RBMK-type reactors was carried out from 1977 to 1986. Graphite stack restoration terms 2013–2019.



Figure 3 ¹⁴C concentration in tree rings.

In all measured samples, the contribution of emissions from the Kursk NPP to the concentration of ${}^{14}C$ in the atmospheric air is visible. The highest annual concentrations are observed during the restoration of graphite stack. This confirms that the main part of the produced ${}^{14}C$ remains in the graphite. There is no correlation between annual electricity generation and ${}^{14}C$ emissions. The annual discharges and dose calculation results are presented in Table 4.

Since it is assumed that a major part of ¹⁴C released from RBMKs is in ¹⁴CO₂ form, considering of others chemical forms of ¹⁴C is not required. The values of annual discharges vary from 0.4 TBq in 2020 (regular operation) to 9.9 TBq in 2018 during the restoration of graphite stack. The results of effective dose calculation have demonstrated that contribution of inhalation to ¹⁴C exposure is about 0.03% while ingestion remains the main way of ¹⁴C influence. However, the contribution of background ¹⁴C to the annual effective dose is more than 90%, except for the period of restoration of the graphite stack. If we use an approach that considers the actual consumption of products in the Kursk region, the values of annual effective doses are more than 2 times lower than with the conservative approach of the IAEA.

The selection of tree rings from specific years was made to cover both the period of normal operation and the period of restoration of the graphite stack. During normal operation, in our opinion, it is not necessary to consider the tree ring of each year since ΔpMC changes insignificantly. Probably the period of restoration of the graphite masonry requires more detailed consideration. However, in our paper, we demonstrated the scale of ¹⁴C release during such process.

| | | Annual effective dose (µSv) | | | | |
|---------------|------------------------|-----------------------------|--------------|----------------------|--|--|
| di Year (1 | Annual discharge | IAEA | New approach | | Background ¹⁴ C contribution to | |
| | (10^{12} Bq) | approach | Ingestion | Inhalation | effective dose (%) | |
| 1976 | 2.8 | 1.39 | 0.67 | $2.24 \cdot 10^{-4}$ | 92.5 | |
| 1980 | 2.6 | 1.05 | 0.51 | $1.67 \cdot 10^{-4}$ | 93.9 | |
| 1984 | 0.9 | 0.43 | 0.21 | $7.00 \cdot 10^{-5}$ | 97.4 | |
| 1986 | 1.1 | 0.48 | 0.23 | $7.90 \cdot 10^{-5}$ | 97.0 | |
| 1988 | 2.4 | 1.13 | 0.55 | $1.86 \cdot 10^{-4}$ | 92.8 | |
| 1990 | 1.9 | 0.88 | 0.43 | $1.46 \cdot 10^{-4}$ | 94.3 | |
| 1994 | 0.6 | 0.57 | 0.28 | $9.59 \cdot 10^{-5}$ | 96.2 | |
| 1998 | 0.4 | 0.35 | 0.17 | $6.10 \cdot 10^{-5}$ | 97.6 | |
| 2002 | 2.0 | 0.40 | 0.19 | $6.96 \cdot 10^{-5}$ | 97.1 | |
| 2006 | 3.7 | 1.86 | 0.90 | $3.34 \cdot 10^{-4}$ | 87.6 | |
| 2010 | 0.8 | 0.66 | 0.32 | $1.21 \cdot 10^{-4}$ | 95.2 | |
| 2014 | 8.3 | 5.30 | 2.58 | $9.89 \cdot 10^{-4}$ | 70.9 | |
| 2018 | 9.9 | 4.58 | 2.23 | $8.75 \cdot 10^{-4}$ | 73.7 | |
| 2020 | 0.4 | 0.17 | 0.08 | $3.20 \cdot 10^{-5}$ | 98.7 | |

Table 4 The values of average annual discharge and effective dose.

CONCLUSION

In this paper, we analyzed the concentration of ¹⁴C in tree rings in the vicinity of the Kursk NPP. Accelerator mass spectrometry measurements demonstrated Δ^{14} C excesses in each of the samples, which are due to the operation of four RBMK reactors: from 1.3 pMC in 2020 (normal operation) to 41.6 pMC in 2014 during the restoration of the graphite stack. The assumption that the main part of ¹⁴C is emitted as ¹⁴CO₂ made it possible to perform a retrospective assessment of annual releases and effective doses. The release value ranged from 0.4 to 9.9 TBq. Accounting for real consumed products in the Kursk region reduced the values of annual effective doses by more than 2 times compared to the IAEA approach: 0.17–5.30 µSv (IAEA) and 0.08–2.58 µSv (new approach). Background contribution is the main part of ¹⁴C exposure—from 70 to 99%

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