

An Analytical Electron Microscopy Study of Au/TiO₂ Thin Films Deposited on the Glass Substrate

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Gold/TiO₂ films have attracted much attention due to the high catalytic activity, and have recently been characterized using surface enhanced Raman scattering (SERS) technique [1]. Red-shift in the extinction spectrum was observed with the increase of TiO₂ film thickness and has ascribed to the increase in effective refractive index of the substrate slab and coupled plasmon resonance. A similar Au/TiO₂ structure was hence investigated in confirming the propriety of the SERS analysis and disclosing the point at issue using analytical electron microscopy.

A TiO₂ thin film was deposited on an optical microscope glass slide (FEA, ground edges, plain) at 200°C, by 417 cycles of atomic layer deposition (ALD) using alternating pulses of tetrakisdimethylamido titanium and H₂O with an Ar carrier gas flow. The deposition rate for TiO₂ was 0.045 nm per ALD cycle [1], and the estimated film thickness was 20 nm using a spectroscopic ellipsometer. An Au layer was then deposited on the surface of TiO₂ film by thermal evaporation. The thickness of the Au layer was ~ 3 nm as indicated by a quartz crystal thickness monitor. The specimen was reinforced with carbon and Pt layers, and then thinned by focused ion beam (FIB). Analytical TEM was performed using a JEM-2100F UHR STEM, equipped with an Oxford Instruments 30 mm² EDS spectrometer, a Gatan Orius model US1000FT CCD camera, and a GIF Tridiem (model) energy filter.

HR-TEM revealed that the 3 nm thick Au film was composed of round particles with diameters of ~15 nm or less [2]. Based on TEM images, the thickness of TiO₂ layer was measured to be 11 nm though the expected thickness was 20 nm (as mentioned above). STEM HAADF image, as shown in Figure 1a, depicts each Au⁷⁹ particle clearly. The Ti²²O⁸ layer (arrow in Fig. 1a) is shown with light (contrast) gray band that is laid in between dark areas of the glass (Si¹⁴O⁸) and bright Au⁷⁹ layers. EFTEM thickness map (Fig. 1b) illustrates t/λ , where λ is the average mean free path of inelastic scattering of electrons and t is the local specimen thickness, which can be evaluated from $t = \lambda \log(I_T/I_0)$ [3]. I_0 is zero-loss intensity, obtained from zero-loss image (0±5 eV) (Fig. 1c), and I_T is the total intensity in EELS, obtained from conventional EFTEM image (Fig. 1d). This thin specimen, with $t < 0.31\lambda$, warrants the accuracy of the EELS analysis. The EFTEM Ti-L map (Fig. 1e) and EFTEM C-K map (Fig. 1f) indicate the TiO₂ layer and the reinforcement of C, respectively. Figure 2 shows EDS analysis of the Au/TiO₂ microstructure. HAADF image (Fig. 2a) and Au-L map (Fig. 2b) reveal an Au particle. The Ti-L map (Fig. 2c) indicates the TiO₂ layer whereas Si-K map (Fig. 2d) indicates the glass substrate. However, these images exhibit artifact contrasts due to the higher background of X-rays stimulated by strong emissions from Au particle. The Ca-K map (Fig. 2e) illustrates Ca atoms on the common soda lime glass substrate. EELS maps (Fig. 3) do not exhibit any artifact in the gold region. The O-K map (Fig. 3a), O-K map with energy-loss near-edge structure for Ti-O (Fig. 3b) and together with Ti-L map (Fig. 3c) confirmed the existence of TiO₂ layer. It is known that the density of amorphous oxide films is ~30 % less than that of the bulk oxide and the amorphous films may contain many voids [4]. As the soda lime glass substrate was heated to 200 °C during ALD, Ca atoms in glass surface have become mobile and readily diffused a

few nanometers into the thin amorphous TiO₂ film (Fig. 3d). Those voids in the thin amorphous TiO₂ layer might have been immediately occupied by diffused Ca atoms while heating the substrate (i.e., before the deposition of Au). Therefore, Au atoms that lately evaporated/deposited on the surface of amorphous Ti(Ca)O₂ could easily migrate around and formed larger Au nuclei. In contrast, the surface of the thicker amorphous TiO₂ layer was consisted of pure TiO₂ with more voids and thus Au particles are confined and formed smaller nuclei compared to a thin amorphous Ti(Ca)O₂ layer. The difference in the number of Au nuclei on different TiO₂ layers may explain the geometrical differences of Au particles that have reported in ref (1).

The diffusion of Ca atoms into TiO₂ film during ALD might have resulted in the followings: 1. the different chemical composition in different thickness of TiO₂ layers especially at the surface of substrate, 2. the different growth behavior of Au particles, and 3. variation of electric and optical properties of the deposited films. Ultra-high resolution analytical microscopy study is, thus, indispensable in the characterization of nanoscale novel materials and devices.

References

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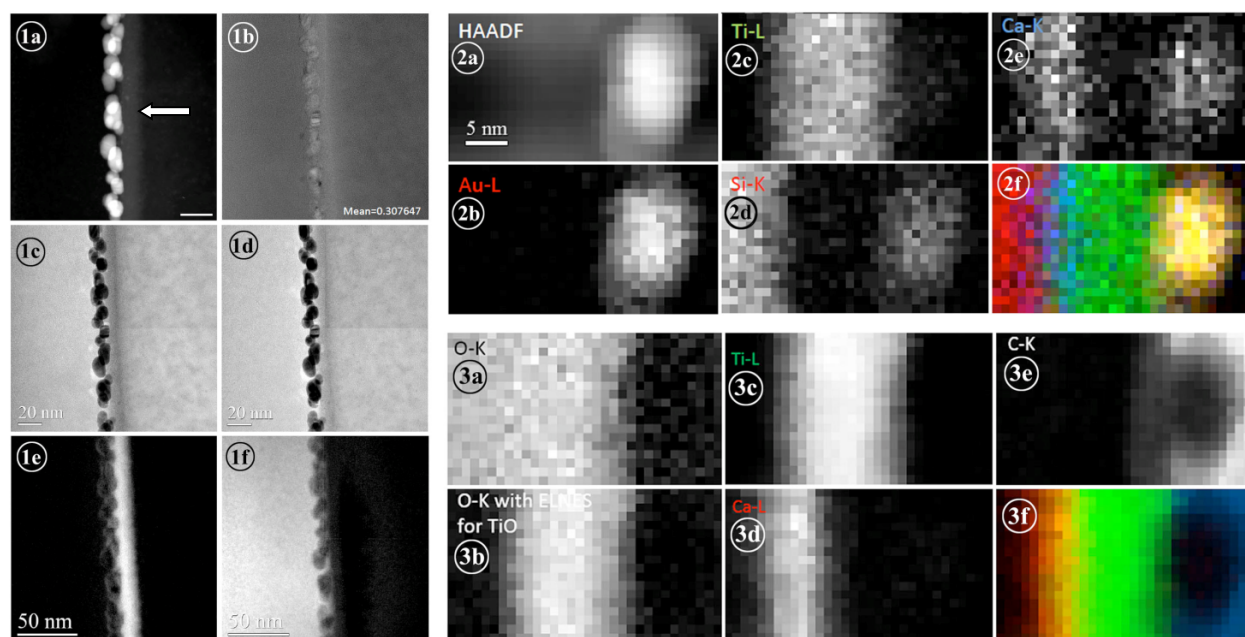


Fig. 1. EFTEM of Au/TiO₂ microstructure. (a) STEM HAADF image. (b) EFTEM thickness map. (c) Zero-loss image (0 ± 5 eV). (d) Conventional EFTEM image (e) EFTEM Ti-L map. (f) EFTEM C-K map.
 Fig. 2. EDS mapping of Au/TiO₂ structure. (a) HAADF image. (b) Au-L map. (c) Ti-L map. (d) Si-K map. (e) Ca-K map. (f) Composite EDS map (Si: red, Ca: blue, Ti: green and Au: yellow).
 Fig. 3. EELS maps of Au/TiO₂ structure. (a) O-K map. (b) O-K map with energy-loss near-edge structure for Ti-O. (c) Ti-L map. (d) Ca-L map. (e) C-K map. (f) Composite EELS map (Ca: red, Ti: green and C: blue).