

Study of the Molybdenum Oxide Rods Sulfidation Process

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Molybdenum sulfide (MoS_2) is widely used as a catalyst, semiconductor, and tribology. MoS_2 has a laminar crystalline structure with Van der Waals forces between S-Mo-S layers in (002) direction. Different molybdenum compounds synthesis methods and morphologies have been reported [1, 2]. Some of these get molybdenum morphologies as oxides, following sulfidation process to obtain MoS_2 [1]. The stability of oxide structures and the formation of intermediate suboxides hinder the sulfidation process of the material [3]. Different conditions were studied to achieve complete molybdenum sulfidation.

Molybdenum oxide (MoO_3) was synthesized by the microwave-assisted hydrothermal method (MAHM) previously reported by Paraguay-Delgado et al [4]. To synthesize MoO_3 , an aqueous solution of ammonium heptamolybdate 0.3 M was heated in NAHM at 200°C and 41 bar for 20 minutes. To obtain MoS_2 , the MoO_3 was treated in a tubular furnace in the sulfur atmosphere. Sulfidation conditions like heating rate, temperature, atmosphere and time of treatment were adjusted to obtain complete MoS_2 . The materials were characterized by TGA-DTA, XRD, SEM, EDS and HRTEM techniques.

Figure 1 shows the XRD of KC obtained by NAHM and MoO_3 after sulfidation process. The MAHM synthesis gives a hexagonal MoO_3 and no other phases were detected. Figure 2 a) shows the SEM images of KC, which present hexagonal rod morphology with 11.9 μm long and 3.3 μm diameter averages. Traditionally sulfidation conditions ($\text{KM-500}^\circ\text{C-H}_2\text{S/H}_2$) resulted in low sulfurized molybdenum oxide with 7.4% of MoS_2 according to EDS analyses, high-temperature sulfidation (700°C) in the same conditions ($\text{KM-700}^\circ\text{C-H}_2\text{S/H}_2$) give just 26.6% of MoS_2 . According to XRD traditionally synthesis at 500°C and 700°C samples have phase change to MoO_2 (figure 2), complete reduction of MoO_3 to MoO_2 was done. Complete sulfuración was achieved at 700°C with $\text{H}_2\text{S/N}_2$ flow ($\text{KM-700- H}_2\text{S/H}_2$) showed in figure 2 d).

References:

- [1] C. Ornelas, F. Paraguay-Delgado and J. Lara-Romero, *J. Mater. Res. Technol* 4 (2019), p. 3672–3680
- [2] P. Kumar, et al, *J. Alloys Compd.*, 671 (2016), p. 440–445.
- [3] T. Leisegang, et al, *Cryst. Res. Technol.* 40 1–2 (2005), p. 95–105.

[4] M. Santos-Beltran, et al, J Mater Sci: Mater Electron 28, I 3 (2017), p. 2935-2948

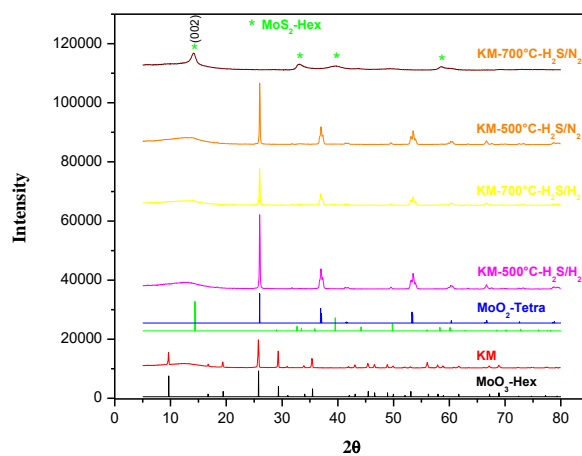


Figure 1. Molybdenum oxide X-ray diffraction.

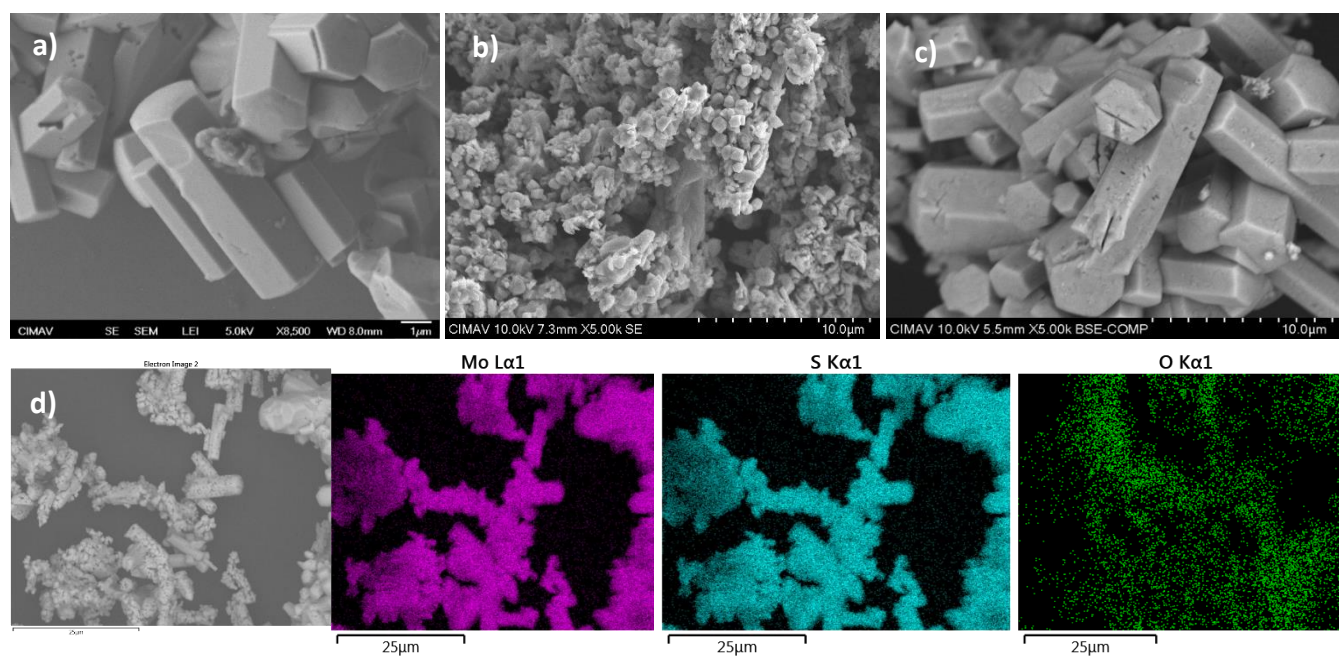


Figure 2. SEM micrographs of a) KM b) KM-700°C-H₂S/H₂ c) KM-700°C-H₂S/N₂ and d) KM-700°C-H₂S/N₂ elemental mapping.