# CrossMark

## Examination of Hydrated Bacteria in An Environmental Scanning Electron Microscope (ESEM)

Ann Fook Yang and Miloslav Kaláb Agriculture and Agri-Food Canada Ottawa, Ontario, Canada K1A 0C6

An ESEM has recently become the tool of choice for many laboratories. Biologists have taken advantage of its capability to examine uncoated fresh specimens or certain types of specimens from insect collections that should not be gold-coated for examination in a conventional scanning electron microscope. However, some fresh specimens are more difficult to examine than others when using an ESEM—specimens consisting of bacteria are particularly difficult. Recently, a question on this subject was posted on the Microscopy Listserver, where it brought many responses. They are summarized below.

It should be noted that pathogens in the specimen chamber can contaminate the whole instrument and be aerosolized into the room. Therefore, unknown bacteria and pathogens must be fixed prior to ESEM examination.

#### General procedures

- Pre-cool the cold stage to a temperature (T) of 2°C, place the untreated bacteria onto it directly or mounted on a pre-cooled cold-stage stub with excess water surrounding the specimen.
- Select the "Wet Specimen" mode, "Target Vapor Pressure" (VP) 5.3
  Torr, "Flood Cycle" (4-6X) above the target VP (5.5 10 Torr). The
  T and VP may be varied as desired as long as it translates to 100%
  relative humidity.
- Place double-distilled water in the corner wells of the cold stage or place
   1-2 ml in a tube positioned in the chamber. Pump out the chamber.
- 4. Work at a low working distance, e.g. 7 mm in the ESEM; reduce the VP very slowly while monitoring the excess water being evaporated. Focus quickly and capture images before the specimen is damaged by the electron beam.

Bacteria on tissue or inorganic matter may be viewed *in situ* by gluing the substrate onto the cold stage stub with colloidal graphite or a double-sided carbon tab. Salts and other organic matter in the culture must be washed away. Therefore, bacteria in liquid media must be centrifuged gently and washed with water. The suspension may be placed on the cold stage directly or filtered with a carbon-coated membrane filter (polycarbonate), which is then glued onto a cold stage stub. For bacteria on agar, it is advisable to cut a small plug, remove excess agar and place it on the cold stage. A TEM grid on the surface of the agar can be used as a focusing aid and for locating the bacteria. However, bacteria usually have a protective polysaccharide layer on the top of their colonies that makes viewing the bacteria difficult or even impossible.

Images of fresh bacteria by ESEM are not as crisp as critical point dried (CPD) and subsequently gold-coated specimens examined by conventional SEM for several reasons:

- Many bacteria are covered with a layer of polysaccharide. This layer is retained in an ESEM where it obscures the underlying cells, in contrast to conventional SEM, where this layer is removed during specimen preparation.
- 2) Gold on CPD specimens yields a stronger secondary electrons (SE) signal than a hydrated specimen can generate in ESEM (using a tungsten gun), and the ESEM signal is from a greater depth resulting in softer images.

Although some biologists fail to get high-quality images of fresh bacteria using an ESEM, there are others who recognize the constraints of the technique and are satisfied with the less crisp ESEM images obtained promptly, if they provide the information at the level anticipated. Field emission ESEM should produce better images than microscopes equipped with W or LaB<sub>6</sub> filaments. The satisfaction with image quality depends, to a great extent, upon the research objective.

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Bill Miller <microbill@mohawk.net>
Frank Thomas <thomasf@gsca.NRCan.gc.ca>
Frederick C. Monson <fmonson@wcupa.edu>

Glenn Poirier <glpoirie@NRCan.gc.ca>
Louis Harner <Louise\_Harner@albint.com>
Scott Whittaker <Whittaker.scott@nmnh.si.edu>
Tony Bruton <Bruton@nu.ac.za>
David Patton <David.Patton@uwe.ac.uk>
Scott Robinson <sjrobin@itg.uiuc.edu
Daniel Phifer <dphifer@feico.com>
Frank Baker <fsb21@phy.cam.ac.uk>
Jose Saleta <saleta@bren.ucsb.edu>

### Dye is Money

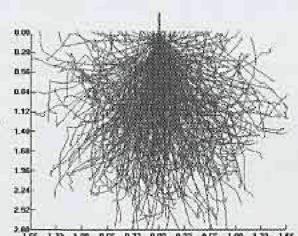
Christian Lohr University of Kaiserslautern, Germany clohr@rhrk.uni-kl.de

In our laboratory, we often use fluorescent dyes in aqueous solution, for example calcium indicators that are iontophoretically injected into neurons through a microelectrode. Usually, the dyes are dissolved in water or electrophysiological recording solution as a stock solution, and aliquots are kept frozen until used. This treatment, however, does not prevent the degradation of the dye over a longer period of time. The potassium salt of Fura Red (Molecular Probes), for instance, looses its sensitivity to calcium after approximately three months of storage in aqueous solution at -20°C. The dry salt, in contrast, can be stored for years when kept dark and frozen. Hence, it is advantageous to store aliquots of the dry salt. The amount of dye, which usually is in the range of 1 mg, is too small to be directly divided into smaller portions. Therefore, we dilute the dye in a small volume of methanol, subdivide the solution into aliquots in 1-ml reaction tubes, and then let the methanol evaporate for 1 hour in darkness. For example, 1 mg of Alexa Fluor 488 hydrazide sodium salt is diluted in 100 µl of methanol, and aliquots of 5 µl are made. After evaporation, the aliquots are kept frozen in darkness and are stable for one or more years. We have successfully tried this method for different Alexa Fluor hydrazide

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salts, the fluorescent calcium channel ligand BODIPY FL verapamil, and the salts of the calcium indicators Fura Red, Fluo-4, and Oregon Green BAPTA-1. It should be noted that measuring solvents like methanol with a pipette can cause slightly imprecise volumes because of the high vapor pressure. In general, this method can also be used to prepare aliquots of drugs like receptor ligands or ion channel blockers.

# Water Recirculator (Chiller) Maintenance<sup>1</sup>

Owen Mills

Michigan Technological Institute opmills@mtu.edu

In my mind two things commonly cause problems for electron microscopes. One is vacuum, and the other is cooling water. It is really not that they are trouble, but rather that extra maintenance is involved. The maintenance of water recirculating systems will be discussed in this note.

A water recirculator is really a heat exchanger that operates on a basic heat transfer principle. The process cooling water within the recirculating system is forced through the scope by a pump. Heat is transferred to the process cooling water as it flows through the microscope lenses, power supplies, and diffusion pump. The process cooling water then flows back through a heat exchanger inside the water recirculator. The total heat is transferred through this heat exchanger into one of two cooling sources:

1. Refrigerated cooling system (evaporator)

2. Non-refrigerated cooling system (water-to-water heat exchanger)

In a refrigerated water recirculating system, the total heat (heat from the microscope plus heat from the mechanical compressor) is transferred through the evaporator by refrigerant into a second heat exchanger known as a condenser. A condenser may be air cooled or it may be water cooled. An air cooled condenser will transfer the total heat into the surrounding ambient air. A water cooled condenser will transfer the total heat into an in-house source of cooling water. The typical temperature range for the condenser water circuit is 50-85°F.

In a non-refrigerated water recirculating system, the total heat is transferred through the water-to-water heat exchanger directly into an in-house source of cooling water. The typical temperature range for the water-to-water heat exchanger circuit is 40-55°F.

Concerning maintenance, the tank water is checked weekly to insure a proper operating level. It is also a good idea to check the pH level of the water to make sure that it remains neutral. When it becomes dirty, it is drained and refilled with clean water. This occurs about every two years. After the system is drained, the intake filter is cleaned or replaced. A large ultrasonic cleaner can be used to clean the intake filters. Since the intake is directly before the pump, and in order to prevent pump failure, the water pump should always be primed by manually filling the intake hose before installing the filter. Distilled water is used in all systems, and mineral buildup has not been a problem. Except for one scope, all the supply hoses are opaque. Algae problems in that scope chiller are treated with Chloramine T (Sigma Chemical), which effectively kills it.

The pumps used on all recirculators usually are connected to their motors via a drive key or pump coupling. This item wears under some conditions, and when it does, debris typically will be found beneath the motor and pump connection. It is a good idea to have a spare motor and pump on hand. In my experience, pumps may last 4 years or more. Few problems have been experienced with the motors themselves. Often, repair shops specialize in the repair of electrical motors. The repair can be made for less than the cost of a new motor.

Maintenance of the refrigeration systems is beyond our means with the rigorous control of refrigerants. There are sight glasses on the systems that can be used to monitor the refrigerant level. After the refrigeration compressor has been running a few minutes, bubbles in the sight glass usually indicate a low refrigerant charge. Contact a competent refrigeration specialist to check for leaks and refill if needed. When using an air-cooled condenser as mentioned above, the scope and chiller should be shut down twice a year, or more depending on the amount of dust in the area, and the condenser coils vacuumed. A restricted condenser coil will impede airflow and prevent proper heat transfer. The other heat transfer methods described previously are dependent on adequate flow of in-house cooling water to the system. Flow indicators are mounted on these lines so it is possible to detect instantly when a problem occurs. With a little extra maintenance, water recirculators will not leave us out of business.

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## EDS Spectral Artifacts / Sum Peaks: A Reminder

Steven S. Hurban

Endicott Interconnect Technologies, Inc. Failure Analysis Laboratory, Endicott, NY shurban@eitny.com

Every once in a while we get involved in an analysis that brings back earlier learned lessons. Here is an example of one that proves theory can be seen in practice. It could prove helpful to new SEM/EDS analysts and also as a reminder to the more experienced.

We were looking for low levels of tin in a silicon dioxide (SiO2) matrix and having great difficulty detecting any tin level. In an attempt to stimulate more X-ray information the EDS detector count rate was increased (larger SEM spot size). This resulted in a small energy peak in the 3.4KeV range. We know that the Sn-L $\alpha$  peak should be located at 3.444keV. In haste, the peak we observed could have been mislabeled either automatically using peak ID routines or manually, as the Sn-L $\alpha$  peak we were trying to identify. However, closer examination of the exact peak location revealed that this peak centroid did not line up at the 3.444keV mark. The peak was obviously at a slightly higher energy than where we would expect to see the Sn-L $\alpha$  peak. Also, any secondary peaking associated with the Sn-L $\alpha$  peak was absent. A Si-K $\alpha$  SUM peak would have a centroid of 2x the Si-K $\alpha$  energy (1.740keV), or 3.480keV. The peak centroid lined up perfectly with this energy value. The Sn-L $\alpha$  / Si-K $\alpha$  SUM peak overlap is a classic example of this artifact.

Increasing the count rate to >15,000 counts per second resulted in approximately 60% dead time—not optimum EDS acquisition parameters¹. These instrument parameters can not be overlooked when collecting spectral data. Analyzing the same SiO₂ rich areas using different instrument parameters (<3000 CPS, approx. 20% dead time), we were able to eliminate the artifactual peak at this location. This was supportive proof that the peak was indeed a Si-Kα SUM peak.

This analysis served as a reminder that there are specific instrument/ specimen conditions that will enhance the occurrence of spectral artifacts. We were also reminded to investigate suspicious AUTO ID results. Knowledge of both the sample and the instrumentation capabilities/ limitations play an integral role in accurate and complete elemental analysis. Even today's systems, that incorporate an efficient pulse pileup rejector, can not separate two pulses closely spaced in time. Overloading the EDS detector, limits it's capability to separate pulses in nearly exact coincidence and results in the formation of a small pulse pileup peak in the EDS spectrum.<sup>2,3</sup>

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