




Article

Enricofrancoite, $\text{KNaCaSi}_4\text{O}_{10}$, a new Ca–K–Na silicate from Somma–Vesuvius volcano, southern Italy

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Abstract

Enricofrancoite (IMA2023–002), ideally $\text{KNaCaSi}_4\text{O}_{10}$, is a new litidionite-group member found as the product of high-temperature alteration of hosting silicates with the enrichment by Cu-bearing fluids at the rock–fumaroles interface related to the 1872 eruption of Somma–Vesuvius volcano, southern Italy. It occurs as euhedral and platy crystals or crusts together with litidionite, tridymite, wollastonite and Al- and Fe-bearing diopside, kamenevite, perovskite, rutile, Ti-rich magnetite and colourless Si-glass. Single crystals of enricofrancoite are transparent colourless or light blue with a vitreous lustre. Mohs hardness is 5.5. D_{meas} is 2.63(3) g/cm^3 and D_{calc} is 2.63 g/cm^3 . The mineral is optically biaxial (–), $\alpha = 1.542(5)$, $\beta = 1.567(5)$, $\gamma = 1.575(5)$; $2V_{(\text{meas})} = 60(2)^\circ$ and $2V_{(\text{calc})} = 58^\circ$. The mean chemical composition (wt.%, electron-microprobe data) is: SiO_2 64.81, Al_2O_3 0.03, TiO_2 0.08, FeO 0.07, MgO 1.71, CaO 10.64, CuO 2.22, Na_2O 8.56, K_2O 11.41, total 99.94. The empirical formula based on 10 O apfu is: $\text{K}_{0.90}\text{Na}_{1.03}(\text{Ca}_{0.71}\text{Mg}_{0.16}\text{Cu}_{0.10})_{\Sigma 0.97}\text{Si}_{4.02}\text{O}_{10}$. The Raman spectrum contains bands at 133, 248, 265, 290, 335, 400, 438, 510, 600, 690 and 1120 cm^{-1} and the wavenumbers of the IR absorption bands are: 424, 470, 492, 530, 600, 630, 690, 750, 788, 970, 1040 and 1160 cm^{-1} . The eight strongest lines of the powder X-ray diffraction pattern are [d , Å (I , %) hkl]: 6.75 (42) 01 $\bar{1}$, 3.65 (20) 11 $\bar{2}$, 3.370 (100) 02 $\bar{2}$, 3.210 (52) 102, 3.051 (18) 111, 3.033 (25) 2 $\bar{1}\bar{2}$, 2.834 (22) 02 $\bar{3}$ and 2.411 (72) 03 $\bar{2}$. Enricofrancoite is triclinic, space group $P\bar{1}$, unit-cell parameters refined from the single-crystal data are $a = 7.0155(4)$ Å, $b = 8.0721(4)$ Å, $c = 10.0275(4)$ Å, $\alpha = 104.420(4)^\circ$, $\beta = 99.764(4)^\circ$, $\gamma = 115.126(5)^\circ$ and $V = 472.74(5)$ Å³. The crystal structure has been refined from single-crystal X-ray diffraction data to $R_1 = 0.035$ on the basis of 2078 independent reflections with $F_o > 4\sigma(F_o)$. Enricofrancoite is an H_2O -free analogue of calcinaksite with 5-coordinated Ca^{2+} at the M site.

Keywords: new mineral; enricofrancoite; litidionite group; crystal structure; Somma–Vesuvius; Italy

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Introduction

Enricofrancoite is a new mineral found in rare litidionite-bearing samples of the Somma–Vesuvius volcano, southern Italy ($40^\circ 49' 17''\text{N}$, $14^\circ 25' 35''\text{E}$). The specimen of enricofrancoite in this study was collected, together with other samples of the same type, in the Royal Mineralogical Museum of Naples (henceforth RMMN, nowadays part of ‘Centro Musei delle Scienze Naturali e Fisiche’) of the University Federico II (Italy) in 1873. In June of that year, a mineral collector found small lapilli covered

by deep-blue crusts with a peculiar glassy to enamelled-like appearance, related to the 1872 AD eruption, at the Vesuvius crater and brought them to Arcangelo Scacchi, director of the Museum at that time.

Enricofrancoite is named in honour of Enrico Franco (1927–2009), professor of Mineralogy at the University of Naples Federico II, Italy. Prof. Franco was the author of many papers regarding mineralogy and made a significant contribution to the field of mineralogy of Somma–Vesuvius volcano, where he also discovered panunzite (the natural counterpart of tetrakalsilite) and chabazite-K (de Gennaro and Franco, 1974; Merlini *et al.*, 1985; Franco and de Gennaro, 1988).

The new mineral and its name (symbol *Enf*) were approved by the Commission of New Minerals, Nomenclature and Classification of the International Mineralogical Association (IMA2023–002, Balassone *et al.*, 2023). The holotype specimen is deposited in

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the systematic collection of the Vesuvian Collection of the RMMN, with the catalogue number 17926 E6457.

Occurrence

The specimen with the new mineral (Fig. 1) belongs to a set of samples from Vesuvius volcano (40°49'17"N, 14°25'35"E), which contains the rare litidionite-bearing assemblage (Pozas *et al.*, 1975; Balassone *et al.*, 2019, 2022). The new mineral occurs in a complex paragenesis composed of litidionite, tridymite, wollastonite and Al- and Fe-bearing diopside, kamenevite, perovskite, rutile, Ti-rich magnetite, and an amorphous phase, in deep blue to white crusts covering metamorphosed lapilli related to the 1872 eruption. Various trace non-silicates are also recorded, and further minor to trace phases will very likely be identified in this mineral assemblage from Vesuvius in our ongoing research on new sample sets.

Titanium- and Pb-bearing litidionite are also reported (Balassone *et al.*, 2022); moreover, in the white crust, the Ca–Na–K-bearing silicate was initially identified as calcinaksite by the above authors. This mineral association is typical of high-temperature alteration processes at the rock–fumaroles interface (Pozas *et al.*, 1975; Balassone *et al.*, 2019, 2022) as a product of very localised high-temperature (exhalation-related) alteration processes of hosting silicates and the introduction of other chemical components by fluids (e.g. Cu²⁺). This peculiar assemblage, like that observed for the Eifel region calcinaksite-bearing rocks (Aksenov *et al.*, 2014; Chukanov *et al.*, 2015), probably occurs at a temperature of >600°C. Further chemical investigations predominantly on the white assemblage revealed a significant mineral component corresponding to an anhydrous analogue of calcinaksite, which was then named enricofrancoite. In the

observed mineral assemblage, there are four different types of litidionite-group minerals which crystallise in the following order: euhedral and platy crystals of litidionite or Ca-bearing litidionite up to enricofrancoite → altered zones around diopside crystals consisting of litidionite or fine-grained litidionite → coarse-grained enricofrancoite crystals in the litidionite matrix → marginal zones of lapilli represented by Ti-bearing non-stoichiometric litidionite (Balassone *et al.*, 2022). Based on our new evidence, most of the calcinaksite indicated previously should be in fact enricofrancoite.

Physical and optical properties

The morphology of the minerals was studied using a Jeol JSM5310 scanning electron microscope with an Oxford EDS equipped with an INCA X-stream pulse processor and the 4.08 version Inca software (Department of Earth Science, Environment and Resources, DiSTAR, University of Naples Federico II Italy). Enricofrancoite forms platy crystals in fine-grained aggregations (Fig. 2), rarely to 1 mm in length. It forms colourless, translucent, non-fluorescent individual crystals, with a white streak and vitreous lustre. The mineral is brittle, with a stepped fracture, a good {001} cleavage. The calculated density is 2.63 g/cm³, from single-crystal X-ray diffraction (SCXRD) data, as the measured density by flotation in Clerici solution, i.e. 2.63(3) g/cm³. The compatibility index is –0.049, good based on empirical formula and measured density (Mandarino, 2007). Mohs hardness is assumed to be 5.5. The new mineral is transparent and optically biaxial (–), the refractive indices were measured in white light: $\alpha = 1.542(5)$, $\beta = 1.567(5)$ and $\gamma = 1.575(5)$. The $2V_{(\text{meas})} = 60(2)^\circ$ and $2V_{(\text{calc})} = 58^\circ$. The $2V$ was measured using the formula $\sin V = KD/\beta$, where D is the distance in conoscopy between the vertices of the hyperbola branches at their maximum divergence (measured with an eyepiece equipped with a micrometre ruler); β is the refractive index; K is a coefficient depending on the optical system of the microscope, which was calculated using a standard with a known value of $2V$ (phlogopite – 16°). The dispersion is weak ($r < v$), and it is non-pleochroic. The optical orientation is $X \approx b$, $Y \wedge c = 49^\circ$ and $Z \wedge a \approx 90^\circ$.

Chemical composition

The chemical composition was determined with a Cameca MS–46 electron microprobe operating in a wavelength-dispersive mode (WDS) at 20 kV and 20–30 nA and a spot size of 10 μm at the Geological Institute of the Kola Science Center, Apatity. The following standards were used: lorenzenite (Na, Ti); pyrope (Mg, Al); diopside (Si, Ca); wadeite (K); hematite (Fe); metallic niobium (Nb); metallic copper (Cu); wulfenite (Pb); and atacamite (Cl). The mean analytical results of three selected crystals (five spots for each crystal, with a total of 15 analyses) are given in Table 1. Cation contents were calculated with the MINAL program of D. Dolivo-Dobrovolsky (Dolivo-Dobrovolsky, 2016). H₂O was not analysed because of the absence of bands corresponding to the O–H vibrations in the Raman and Fourier-transform infrared spectroscopy (FTIR) spectra.

The empirical formula calculated on the basis of O = 10 atoms per formula unit (apfu) is $\text{K}_{0.90}\text{Na}_{1.03}(\text{Ca}_{0.71}\text{Mg}_{0.16}\text{Cu}_{0.10})_{\Sigma 0.97}\text{Si}_{4.02}\text{O}_{10}$. The analyses of enricofrancoite crystals reveal a substitution between Ca²⁺, Mg²⁺ and Cu²⁺ nearly equal to 0.1 apfu. It seems that the Cu–Ca substitution forms a continuous series between litidionite and enricofrancoite. The amount of Mg is also variable and it is possible that the Mg-analogue of

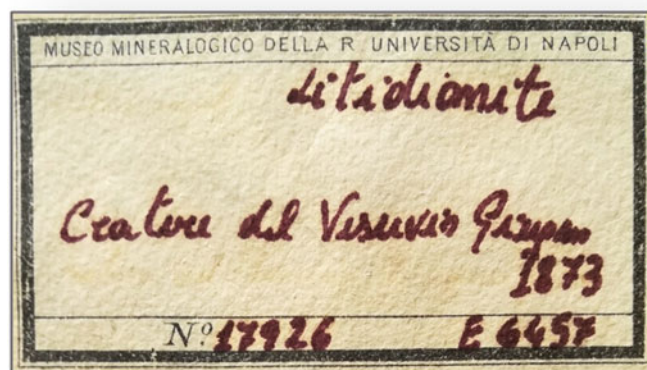
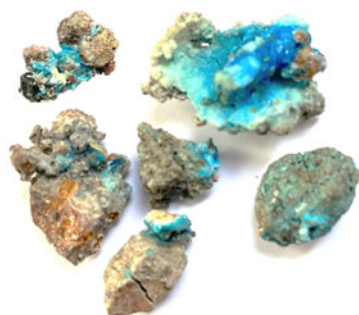


Figure 1. The enricofrancoite-bearing sample (holotype #17926 E6457) from the Royal Mineralogical Museum of Naples University Federico II (modified after Balassone *et al.*, 2022). The length of the Museum label is 7 cm.

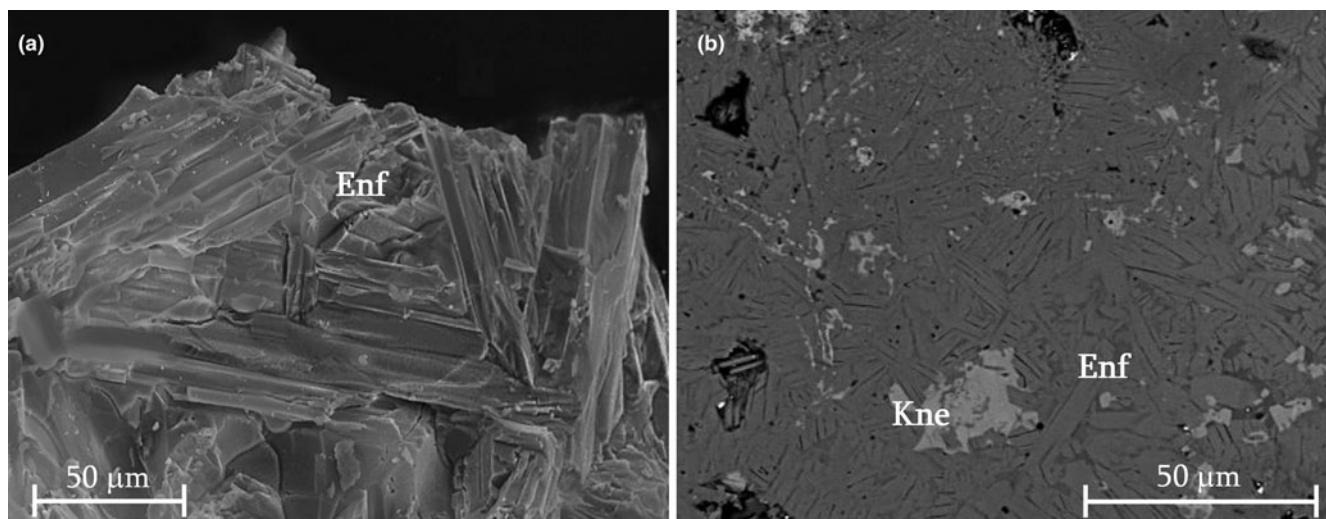


Figure 2. Scanning electron micrograph of (a) enricofrancoite (Enf) in aggregations of platy crystals (holotype #17926 E6457) and (b) back-scattered electron mode image of enricofrancoite with kamenevite (Kne) in polished thin section.

enricofrancoite is also present in the same mineral association. According to our previous data, the Ti^{4+} (analysis No 2 in Table 1) is also incorporated into the *M* site together with the Mg, Cu and Fe. This substitution may also include K/vacancy variations in the composition (Balassone *et al.*, 2022). The simplified formula is $KNa(Ca,Mg,Cu)Si_4O_{10}$. The ideal formula is $KNaCaSi_4O_{10}$, which requires K_2O 12.58%, CaO 14.97%, Na_2O 8.28%, SiO_2 64.18%.

Crystal structure solution and refinement

The grain selected for optical study was also used for single-crystal X-ray diffraction. The crystal studied was mounted on an Oxford Diffraction Xcalibur Eos diffractometer equipped with a CCD area detector using $MoK\alpha$ radiation (0.71073 Å) at the X-ray Diffraction Resource Centre of St. Petersburg State University. More than a half sphere of the diffraction sphere was collected (scanning step 1° , exposure time 75 s). The data were integrated and corrected by means of the *CrysAlisPro* program package,

Table 1. Chemical data (in wt. %) for enricofrancoite.

wt.%	1	2	3	Reference material
SiO_2	66.35	64.15	63.92	Wollastonite
Al_2O_3	0.02	0.05	0.03	Pyrope
TiO_2	0.08	0.12	0.04	Lorenzenite
FeO	0.08	0.09	0.05	Hematite
MgO	1.52	1.55	2.06	Pyrope
CaO	11.37	10.62	9.94	Wollastonite
CuO	2.07	1.60	3.00	Copper
Na_2O	8.98	8.52	8.17	Lorenzenite
K_2O	10.35	11.82	12.06	Wadeite
Total	100.82	98.53	99.28	
Calculated on the basis of $O = 10$ apfu				
Si^{4+}	4.04	4.03	4.00	
Al^{3+}	-	-	-	
Ti^{4+}	-	0.01	-	
Fe^{2+}	-	-	-	
Mg^{2+}	0.14	0.14	0.19	
Ca^{2+}	0.74	0.71	0.67	
Cu^{2+}	0.10	0.08	0.14	
Na^+	1.06	1.04	0.99	
K^+	0.80	0.95	0.96	

which was also used to apply an empirical absorption correction using spherical harmonics, as implemented in the *SCALE3 ABSPACK* scaling algorithm (Agilent Technologies, 2014). The crystal structure was drawn using the *VESTA 3* program (Momma and Izumi, 2011). The distortion indexes for polyhedra were calculated according to the formula proposed by Baur, (1974).

As the calcinaksite and enricofrancoite are chemically very close, the structure of enricofrancoite was refined using the initial calcinaksite model (Chukanov *et al.*, 2015) without O11 (Ow) site to $R_1 = 0.035$ for 2078 independent reflections with $F_o > 4\sigma(F_o)$ using *Olex2* (Dolomanov *et al.*, 2009) with *SHELXL* (Sheldrick, 2015). Initially the *M* site refined with full occupancy by Ca atoms only, but the Bond-valence analysis demonstrated an over-estimated sum (2.45 valence units), which was probably connected to admixtures of Mg and Cu with smaller ionic radii than Ca. According to the EMPA data the composition of the *M* site should be $(Ca_{0.71}Mg_{0.16}Cu_{0.10})_{0.97}$. At the final stage of refinement, the *M1* site occupancy were fixed with $(Ca_{0.71}Mg_{0.19}Cu_{0.10})_{1.00}$. The R_1 factor slightly improved from 0.0353 in a model with $Ca_{1.00}$ occupancy to 0.0348 for the last data with mixed occupancy. The SCXRD data are deposited in the CCDC under the entry No. 2292985 [Cambridge Crystallographic Data Centre, <https://www.ccdc.cam.ac.uk/>] and crystallographic information files are available as Supplementary material here (see below). Crystal data, data collection information and structure refinement details are given in Table 2. Atom coordinates, anisotropic atomic displacement parameters, and selected interatomic distances are presented in Tables 3, 4 and 5, respectively.

Powder X-ray diffraction

Powder X-ray diffraction (PXRD) data were collected using a Seifert-GE diffractometer ID 3003; the intensity profiles were collected in the 2θ range of $3-80^\circ$, using Ni-filtered $CuK\alpha$ radiation at 40 kV and 30 mA, with a step size 0.02° , at a scanning time of 10 s/step. The diffraction patterns were processed using the *RayfleX* software package. Measurements on the mineral assemblages were also carried out by an automated Rigaku RINT2500 rotating anode laboratory diffractometer (50 kV and 200 mA) equipped with the silicon strip Rigaku D/teX Ultra detector,

Table 2. Crystal data, data collection information and structure refinement details for enricofrancoite.

Crystal data	
Formula	KNaCaSi ₄ O ₁₀
Temperature (K)	293(2)
Crystal system, space group	Triclinic, <i>P</i> $\bar{1}$
<i>a</i> , <i>b</i> , <i>c</i> (Å)	7.0155(4), 8.0721(4), 10.0275(4)
α , β , γ (°)	104.420(4), 99.764(4), 115.126(5)
Volume (Å ³)	472.74(5)
<i>Z</i>	2
ρ_{calc} (g/cm ³)	2.627
μ (mm ⁻¹)	1.775
<i>F</i> (000)	371.0
Data collection	
Crystal size (mm)	0.12 × 0.11 × 0.05
Radiation	MoK α (λ = 0.71073)
2 θ range for data collection (°)	5.974 to 58.616 (full completeness for 50.5° and 80% for 58.6°)
Index ranges	-8 ≤ <i>h</i> ≤ 9, -10 ≤ <i>k</i> ≤ 10, -13 ≤ <i>l</i> ≤ 12
Reflections collected	4065
Independent reflections	2078 [<i>R</i> _{int} = 0.0146, <i>R</i> _{sigma} = 0.0263]
Refinement	
Data/restraints/parameters	2078/0/154
Goodness-of-fit on <i>F</i> ²	1.125
Final <i>R</i> indexes [<i>I</i> > 2 σ (<i>I</i>)]	<i>R</i> ₁ = 0.0348, <i>wR</i> ₂ = 0.0826*
Final <i>R</i> indexes [all data]	<i>R</i> ₁ = 0.0395, <i>wR</i> ₂ = 0.0854
Largest diff. peak/hole (e ⁻ Å ⁻³)	0.73/-0.53

*Weighting scheme: *a* = 0.0299 and *b* = 1.0128

and an asymmetric Johansson Ge (111) crystal (to select the monochromatic CuK α 1 radiation with λ = 1.54056 Å) and using a glass capillary (0.5 mm diameter). The intensity values *I*_{calc} for enricofrancoite were calculated using the VESTA 3 program from single-crystal X-ray data. They are given in Table 6 with the corresponding intensities *I*_{meas} measured from the PXRD data using the software *Analyze RayfleX* (ver. 2.3.3.7, GE Inspection Technologies, USA–Germany), in addition to the calculated (*d*_{calc}) and measured (*d*_{meas}) interplanar distance values and Miller indices (*hkl*). The unit-cell parameters were calculated using UNITCELL software (Holland and Redfern, 1997). The unit-cell parameters refined from the PXRD data are as follows: *a* = 7.071(2) Å, *b* = 8.089(2) Å, *c* = 9.933(2) Å, α = 104.35(2)°, β = 99.51(2)°, γ = 115.28(1)°, *V* = 473.3(1) Å³, space group *P* $\bar{1}$ (#2) and *Z* = 2, which are in good agreement with the single-crystal X-ray diffraction data (Table 2).

Table 4. Anisotropic atomic displacement parameters (Å²) for enricofrancoite.

Atom	<i>U</i> ¹¹	<i>U</i> ²²	<i>U</i> ³³	<i>U</i> ²³	<i>U</i> ¹³	<i>U</i> ¹²
Ca1	0.0128(3)	0.0206(3)	0.0161(3)	0.0054(2)	0.0033(2)	0.0101(2)
K1	0.0298(4)	0.0201(4)	0.0333(4)	0.0054(3)	0.0078(3)	0.0086(3)
Si1	0.0122(4)	0.0166(4)	0.0372(5)	0.0028(4)	0.0090(4)	0.0076(3)
Si3	0.0139(4)	0.0127(4)	0.0168(4)	0.0071(3)	0.0063(3)	0.0064(3)
Si2	0.0221(5)	0.0166(4)	0.0156(4)	0.0060(3)	0.0027(3)	0.0075(3)
Si4	0.0103(4)	0.0142(4)	0.0346(5)	0.0012(4)	0.0044(4)	0.0062(3)
Na1	0.0255(7)	0.0344(8)	0.0225(7)	0.0086(6)	0.0118(5)	0.0174(6)
O1	0.0108(11)	0.0292(14)	0.0388(14)	-0.0109(11)	0.0002(10)	0.0095(10)
O2	0.0387(15)	0.0181(12)	0.0256(13)	0.0036(10)	-0.0035(11)	0.0083(11)
O3	0.0364(15)	0.0233(14)	0.086(2)	0.0234(14)	0.0390(16)	0.0210(12)
O4	0.0285(14)	0.0381(15)	0.0410(15)	0.0096(12)	0.0112(12)	0.0224(12)
O5	0.104(3)	0.0303(16)	0.0172(13)	0.0097(12)	0.0026(15)	0.0080(17)
O6	0.0382(14)	0.0220(12)	0.0268(12)	0.0154(10)	0.0168(11)	0.0156(11)
O7	0.0138(12)	0.0277(15)	0.104(3)	0.0216(16)	0.0118(15)	0.0087(11)
O8	0.0177(12)	0.0280(13)	0.0321(13)	0.0019(11)	0.0046(10)	0.0081(10)
O9	0.0170(11)	0.0226(12)	0.0278(12)	0.0043(10)	0.0004(9)	0.0081(10)
O10	0.0337(14)	0.0332(14)	0.0321(13)	0.0079(11)	0.0183(11)	0.0202(12)

Table 3. Fractional atom coordinates, site occupancy factors (s.o.f.) and equivalent isotropic atomic displacement parameters *U*_{eq} (Å²) for enricofrancoite.

Atom	s.o.f.	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{eq}
Ca1 (=M)	Ca _{0.71} Mg _{0.19} Cu _{0.10}	0.33953(10)	0.88224(9)	0.09591(7)	0.0170(2)
K1	K ₁₋₀₀	-0.14661(13)	0.20386(11)	0.50851(9)	0.0304(2)
Si1	Si ₁₋₀₀	0.08059(14)	0.37154(13)	0.22533(11)	0.0231(2)
Si2	Si ₁₋₀₀	0.55298(13)	0.71354(12)	0.29204(9)	0.0140(2)
Si3	Si ₁₋₀₀	0.30959(15)	1.18447(13)	0.36887(9)	0.0193(2)
Si4	Si ₁₋₀₀	0.35139(14)	0.62847(13)	-0.23022(11)	0.0216(2)
Na1	Na ₁₋₀₀	0.8437(2)	0.8717(2)	0.09196(14)	0.0262(3)
O1	O ₁₋₀₀	-0.0912(4)	0.4316(4)	0.2871(3)	0.0324(6)
O2	O ₁₋₀₀	0.1948(4)	0.9661(4)	0.2620(3)	0.0325(6)
O3	O ₁₋₀₀	0.3501(5)	0.4270(4)	-0.2276(4)	0.0424(8)
O4	O ₁₋₀₀	0.1633(4)	1.2902(4)	0.3436(3)	0.0345(6)
O5	O ₁₋₀₀	0.3429(7)	1.1972(5)	0.5346(3)	0.0625(9)
O6	O ₁₋₀₀	0.6175(4)	0.8887(4)	0.2362(3)	0.0264(5)
O7	O ₁₋₀₀	0.2880(4)	0.5814(4)	0.2480(4)	0.0504(9)
O8	O ₁₋₀₀	0.5544(4)	1.3185(4)	0.3606(3)	0.0296(6)
O9	O ₁₋₀₀	0.4848(4)	0.8067(3)	-0.0807(3)	0.0251(5)
O10	O ₁₋₀₀	-0.0244(4)	0.2188(4)	0.0661(3)	0.0312(6)

Structure description

The crystal structure of enricofrancoite is chemically close to that of litidionite and consists of complex heteropolyhedral framework (Fig. 3a). The main feature of litidionite-related crystal structures is one-dimensional infinite anionic tubes [Si₈O₂₀]⁸⁻ running along [100] (Fig. 3b). Enricofrancoite should be classified as loop-branched dreier double chain {*IB*, 2_∞¹} [³Si₈O₂₀] in accordance with the Liebau classification (Liebau, 1985) or ³*T*₈ silicate tube-containing mineral according to the structure hierarchy for silicate minerals (Day and Hawthorne, 2020). Tubes with the same topology are also observed in the agrellite structure, although they have differences. There is also a ³*T*₈ ribbon in the titanosilicate narsarsukite, which has a different structure and crystal system. Each tube consists of four symmetrically independent Si tetrahedra with Si–O bond distances in the range 1.567–1.634 Å. The mean bond lengths for the <Si1–O>, <Si2–O>, <Si3–O> and <Si4–O> tetrahedra are 1.614, 1.599, 1.605 and 1.617 Å, respectively, in agreement with the full occupancy of all tetrahedral sites by Si atoms.

Table 5. Selected interatomic distances (Å) for the crystal structure of enricofrancoite.

Ca1-O9	2.339(3)	Si1-O10	1.569(3)
Ca1-O9	2.269(2)	Si1-O4	1.629(3)
Ca1-O6	2.175(3)	Si1-O1	1.634(2)
Ca1-O2	2.204(3)	Si1-O7	1.625(3)
Ca1-O10	2.186(3)	<Si1-O>	1.614
<Ca1-O>	2.235		
		Si2-O6	1.567(2)
K1-O6	2.846(3)	Si2-O3	1.615(3)
K1-O2	2.955(3)	Si2-O7	1.610(3)
K1-O8	2.955(3)	Si2-O5	1.604(3)
K1-O4	2.907(3)	<Si2-O>	1.599
K1-O1	3.190(3)		
K1-O1	2.704(2)	Si3-O2	1.570(2)
K1-O7	3.182(4)	Si3-O8	1.625(3)
K1-O5	2.807(3)	Si3-O4	1.615(3)
<K1-O>	2.943	Si3-O5	1.609(3)
		<Si3-O>	1.605
Na1-O9	2.556(3)		
Na1-O6	2.338(3)	Si4-O9	1.581(2)
Na1-O2	2.433(3)	Si4-O8	1.631(3)
Na1-O10	2.648(3)	Si4-O1	1.628(2)
Na1-O10	2.360(3)	Si4-O3	1.629(3)
Na1-O3	2.997(3)	<Si4-O>	1.617
<Na1-O>	2.555		

Note: Ca1 = *M* site

Silicate tubes are connected to each other through the $[\text{Ca}_2\text{O}_8]^{12-}$ dimers, which form layers along the *ab* plane (Fig. 3c) and build a three-dimensional framework (Fig. 3d) (Golovachev *et al.*, 1970; Kornev *et al.*, 1972; Pozas *et al.*, 1975). The square-pyramidal sites form $[\text{Ca}_2\text{O}_8]$ dimers (Fig. 3d) with the Ca–Ca distance of 3.469 Å. In contrast to calcinaksite where Ca atoms are 6-coordinated, in the investigated sample the Ca atoms are 5-coordinated. The 5-coordinated Ca is exceptionally rare in inorganic compounds (Katz *et al.*, 1996). The Ca atom

Table 6. PXRD data for enricofrancoite.*

l_{meas}	l_{calc}	d_{meas} (Å)	d_{calc} (Å)	<i>h k l</i>
42	70	6.75	6.75	0 1 $\bar{1}$
16	32	4.48	4.47	1 0 1
17	6	4.04	4.03	1 1 $\bar{1}$
20	11	3.65	3.65	1 1 $\bar{2}$
100	100	3.370	3.375	0 2 $\bar{2}$
52	38	3.210	3.209	1 0 2
18	22	3.051	3.063	1 1 1
25	85	3.033	3.035	2 1 $\bar{2}$
7	6	2.980	2.986	2 $\bar{2}$ $\bar{1}$
22	34	2.834	2.832	0 2 $\bar{3}$
3	3	2.674	2.675	1 2 $\bar{2}$
11	12	2.566	2.565	2 $\bar{1}$ 2
4	2	2.463	2.462	1 1 2
72	66	2.411	2.411	0 3 $\bar{2}$
17	20	1.9760	1.9799	2 2 $\bar{1}$
2	2	1.9240	1.9245	3 $\bar{3}$ 2
4	4	1.9130	1.9120	3 $\bar{3}$ 2
3	2	1.8200	1.8182	2 $\bar{3}$ 4
3	4	1.7850	1.7870	0 4 $\bar{3}$

*The intensities, measured l_{meas} (from PXRD data) and calculated l_{calc} (from single-crystal X-ray data), are given with the corresponding interplanar distance values (d_{meas} and d_{calc}) and Miller indices (*hkl*). The eight strongest lines are highlighted in bold.

occupies the centre of a distorted square pyramid with five bonds in the range 2.175–2.339 Å (Fig. 3d). In contrast to litidionite or Ti-bearing litidionite with a Cu^{2+} cation at the *M* site, in enricofrancoite there is no strong Jahn–Teller distortion observed. There are two relatively short bonds for Ca–O of 2.187 and 2.175 Å and three bonds with more appropriate distances of 2.204, 2.269 and 2.339 Å. The mean bond distance is 2.235 Å. The refined occupancy of the *M* site ($\text{Ca}_{0.71}\text{Mg}_{0.19}\text{Cu}_{0.10}$)_{1.00} agrees well with its bond valence sum (BVS) of 2.08 vu (Table 7).

The heteropolyhedral framework of enricofrancoite is characterised by a two-dimensional system of crossing channels running

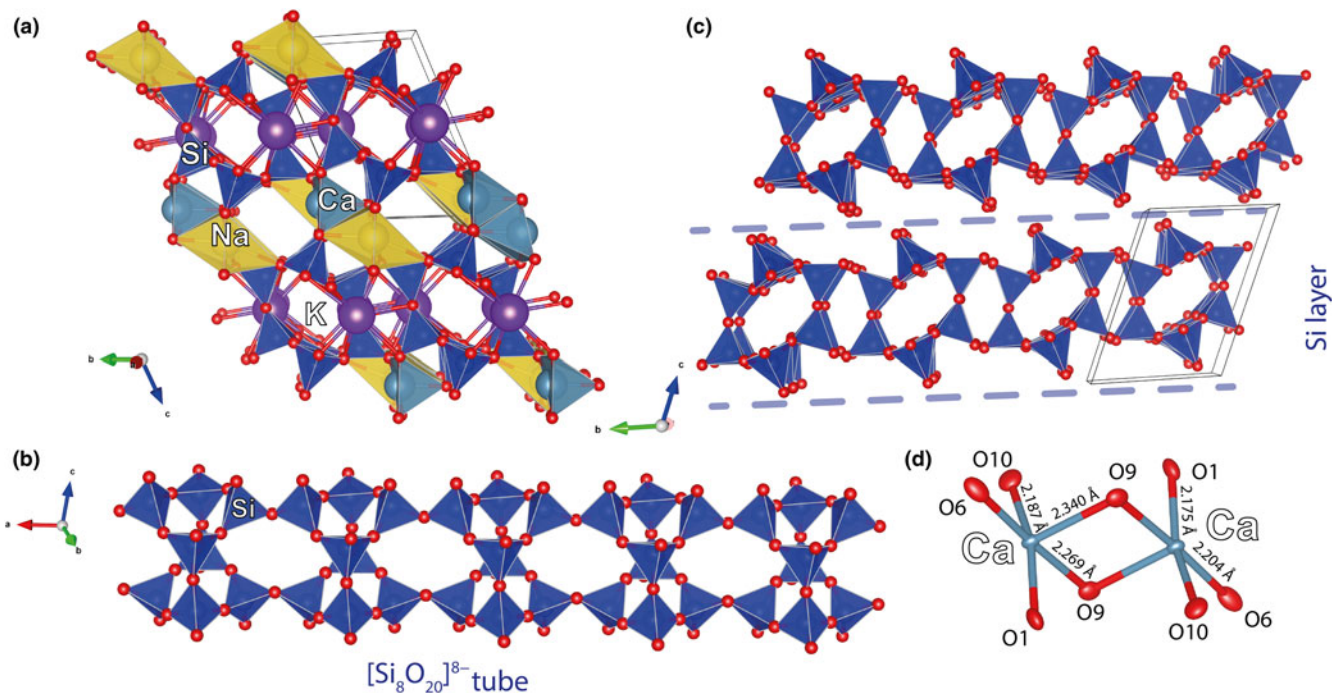
**Figure 3.** Crystal structure of enricofrancoite: (a) general view; (b) silicate tubes; (c) silicate layers; and (d) local coordination of Ca1 site (= *M*) site.

Table 7. Bond-valence analysis (in valence units, v.u.) in the crystal structure of enricofrancoite calculated using bond valence parameters from Brese and O'Keeffe (1991).

Atom	Ca1*	Si1	Si2	Si3	Si4	K1	Na1	Σ
O1		0.97			0.99	0.06 0.21		2.23
O2	0.45			1.16		0.11	0.18	1.90
O3			1.02		0.99		0.04	2.05
O4		0.99		1.02		0.12		2.13
O5			1.06	1.04		0.16		2.26
O6	0.48		1.17			0.14	0.23	2.02
O7		1.00	1.04			0.06		2.09
O8				1.00	0.98	0.11		2.09
O9	0.31 0.37				1.12		0.13	1.93
O10	0.47	1.16					0.10 0.22	1.85
Σ	2.08	4.12	4.29	4.22	4.08	0.97	0.90	

*The BVS for the M site calculated based on the occupancy $[\text{Ca}_{0.71}\text{Mg}_{0.19}\text{Cu}_{0.10}]_{1.00}$.

along the [110] and near parallel to [010] directions of the triclinic cell. For non-isometric channels with an elliptical cross-section, according to International Union of Pure and Applied Chemistry (IUPAC) nomenclature, the lengths of the major and minor axes subtracting the ionic radii of O^{2-} of 2.7 Å are used (McCusker *et al.*, 2003). Channel I (Fig. 4a), running along [010], is characterised by an octagonal cross-section and an effective diameter equal to $4.31 \text{ \AA} \times 1.26 \text{ \AA}$.

Channel II (Fig. 4b) parallel to the [110] direction also has an octagonal cross-section with an effective diameter of $2.88 \text{ \AA} \times 1.87 \text{ \AA}$. Channels I and II overlap and form a two-dimensional system of cavities populated by K^+ cations. The K atom is 8-coordinated with the mean $\langle \text{K}-\text{O} \rangle$ distance of 2.943 Å.

Channel III is composed of four SiO_4 tetrahedra and two CaO_5 pyramids. Channel III runs along the [110] direction and has a hexagonal cross-section and an effective diameter of $1.61 \text{ \AA} \times 1.16 \text{ \AA}$. This type of channel is filled by Na^+ cations. The Na1 site in the enricofrancoite structure is 6-coordinated with a mean $\langle \text{Na}-\text{O} \rangle$ distance of 2.555 Å. Note the weak character of the Na–O3 bond with a distance of 2.997 Å, which gives only (0.04 vu) according to the BVS data. Also according to the BVS data (Table 7), all O sites are fully populated by O^{2-} .

The formula refined from SCXRD data can be written as $\text{KNa}(\text{Ca}_{0.71}\text{Mg}_{0.19}\text{Cu}_{0.10})_{1.00}(\text{Si}_4\text{O}_{10})$.

Raman and FTIR

The Raman spectra of enricofrancoite were recorded with a Horiba Jobin-Yvon LabRAM HR800 spectrometer equipped

with an Olympus BX-41 microscope in back-scattering geometry via a CCD detector. For the measurements uncoated polished sections from the same grain used for EPMA were used. Raman spectra were recorded by a 514 nm Ar laser with power of 2 mW under the 50× objective with a numerical aperture equal to 0.75. The spectra were obtained in the range of 40–4000 cm^{-1} at a resolution of 2 cm^{-1} at room temperature with a 150 μm aperture diameter and 240 s acquisition time. To improve the signal-to-noise ratio, the number of acquisitions was set to 6. For the control of damage to the sample, photos of the sample surface before and after analysis were made. The spectra were processed using *Labspec* (Horiba Jobin Yvon) and *Origin* (OriginLab Corporation) software. The attribution of observed bands was made according to other minerals of the litidionite group (Balassone *et al.*, 2022).

Fourier-transform infrared spectroscopy, performed on a powdered sample in Attenuated Total Reflectance mode (ATR–FTIR), was used in the spectral range 4000–400 cm^{-1} with resolution 2 cm^{-1} (Bruker Alpha; Opus 7.2 software–Bruker Optik GmbH, Leipzig, Germany).

The Raman spectrum of enricofrancoite (Fig. 5) is similar to both Vesuvius litidionite sample R130088 of the RRUFF project (Lafuente *et al.*, 2015) and Ti-bearing litidionite (Balassone *et al.*, 2022), but has some differences. The spectrum band (s = strong band, w = weak band, sh = shoulder) at 1120 (s), can be assigned to the symmetric stretching vibrations of the Si–O bands in the $[\text{Si}_8\text{O}_{20}]^{8-}$ groups (Yakovenchuk *et al.*, 2019; Balassone *et al.*, 2022). Bands at 690 and 600 (s) cm^{-1} related to asymmetric bending vibrations of Si–O–Si and O–Si–O bonds. The bands centred at 510, 438 (w), 400 and 335 (w) cm^{-1} are attributed to different modes of bending vibrations in SiO_4 tetrahedra (Pakhomovsky *et al.*, 2018). Bands at 248, 265 and 290 cm^{-1} can be assigned to symmetric vibrations of Ca–O bonds (Galuskin *et al.*, 2012). The band at 133 (w) cm^{-1} corresponds to the lattice mode vibrations. The absence of bands in the ranges of 1550 to 1650 and 3000–3700 cm^{-1} , respectively corresponding to the bending H–O–H and stretching vibrations of bonds in water molecules and OH-groups, is in accordance with undetectable H_2O content in the enricofrancoite composition.

The IR spectrum of enricofrancoite is given in Fig. 6. In general, the spectrum is somewhat similar to those of the structurally related calcinaksite, fenaksite and manaksite (Chukanov *et al.*, 2015). Wavenumbers of absorption bands can be assigned with Si–O–Si bending and stretching vibrations combined with vibrations of Ca–O bonds in CaO_5 polyhedra 424 (s), 470 (sh) and 492 (sh) cm^{-1} . The bands at 530, 600, 630 (w), 690, 750 and 788 (w) cm^{-1} can be attributed to the bending vibrations of Si–O bonds in the tubular silicate $[\text{Si}_8\text{O}_{20}]^{8-}$ anionic radical.

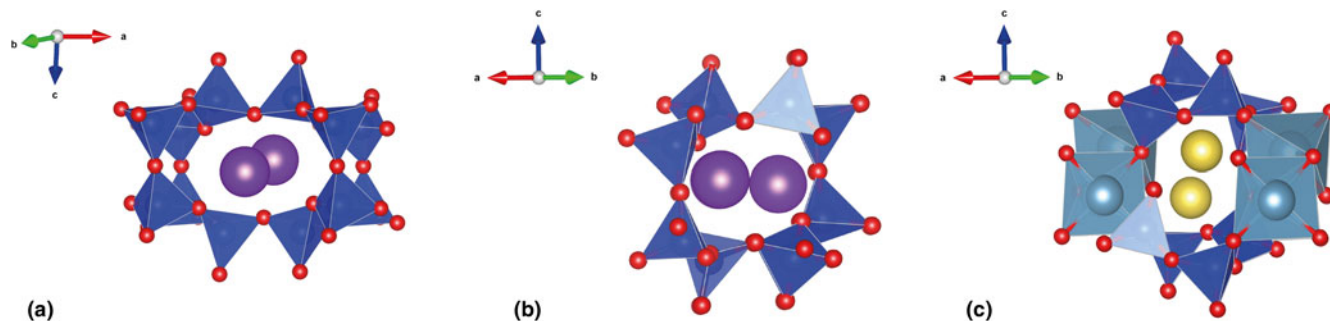


Figure 4. Projections of the channels in the crystal structure of enricofrancoite: (a) along the [010] direction; (b) and (c) two channels along the [110] direction.

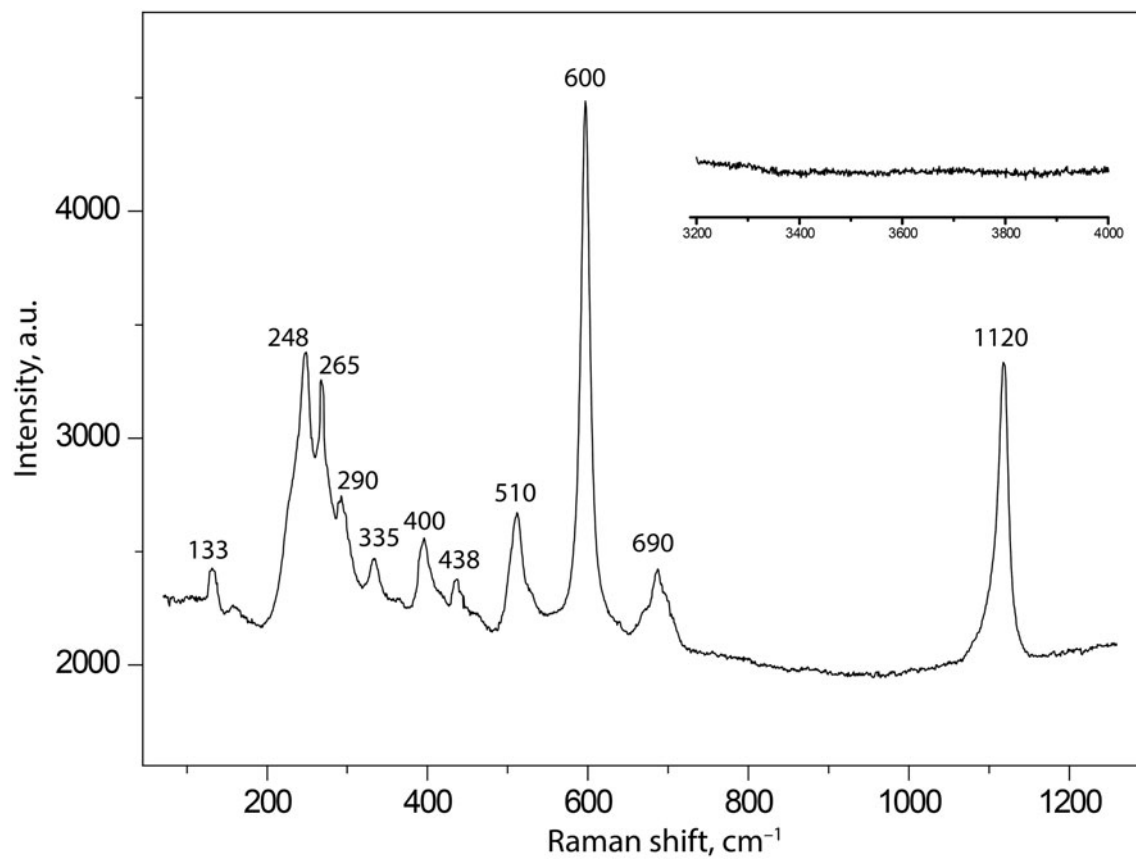


Figure 5. Raman spectrum of enicofrancoite.

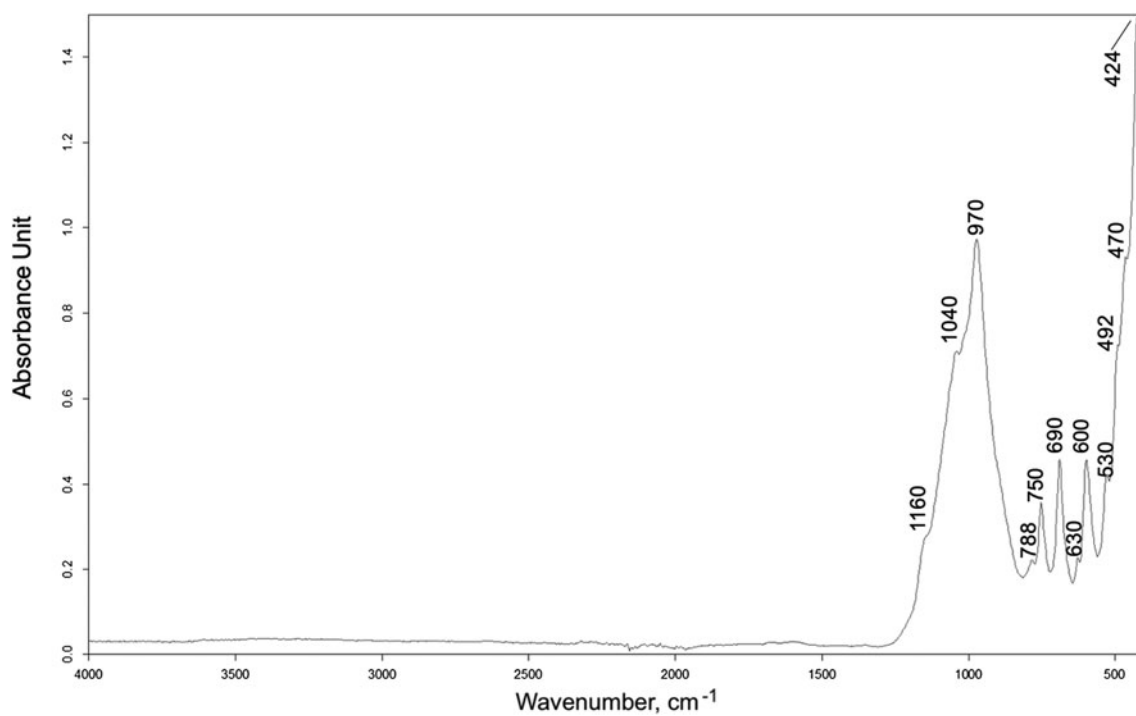


Figure 6. FTIR spectrum of enicofrancoite.

Table 8. Dominant components at the crystallographic sites for selected litidionite-group minerals.

Mineral	Na1	K1	M	Ow (O11)*	Ref.**
Litidionite	Na	K	Cu ²⁺	□	[1]
Fenaksite	Na	K	Fe ²⁺	□	[2]
Manaksite	Na	K	Mn ²⁺	□	[3]
Calcinaksite	Na	K	Ca	H ₂ O	[4]
Enricofrancoite	Na	K	Ca	□	[5]

*The Ow (O11) coordinates are at the Ca1 (=M) site and are only observed in the crystal structure of calcinaksite in which Ca is in octahedral coordination (5+1).

**References: [1] Pozas *et al.* (1975); [2] Golovachev *et al.* (1970); [3] Khomyakov *et al.* (1992); [4] Chukanov *et al.* (2015); and [5] this work.

The bands at 970 (s), 1040 (w), 1160 (sh) correspond to the Si–O stretching vibrations. Bending and stretching vibrations of H₂O molecules have not been observed in contrast to the calcinaksite, which contains certain bands at 1654, 3170, 3340 and 3540 cm⁻¹ (Chukanov *et al.*, 2015).

Discussion

Silicates with tubular fragments, natural and synthetic, are quite rare. In addition to the litidionite family with hexagonal tubes, minerals with octagonal (frankamenite, canasite and miserite) and quadrangular (narsarsukite) tubes, heterogeneous (charoite) and heteropolyhedral tubular fragments (yuksporite) and such compounds often have pronounced magnetic, luminescent and ion-exchange properties (Rainho *et al.*, 2000; Brandão *et al.*, 2009; Day and Hawthorne, 2020). Enricofrancoite is the H₂O-free analogue of calcinaksite with 5-coordinated Ca²⁺ at the M site (also indicated as Ca1). The dominant components in different litidionite-group minerals are listed in Table 8. The most variable site in the litidionite-type structure is the M position, its coordination number can vary from 4+1 (for Cu) and 5 (for Mn²⁺, Fe²⁺, Ti and Ca) to 5+1 (Ca) (Fig. 7a) (Rozhdestvenskaya *et al.*, 2004; Karimova and Burns, 2008;

Brandão *et al.*, 2009; Aksenov *et al.*, 2014; Balassone *et al.*, 2022). A similar diversity for a basically 5-coordinated site is observed for another silicate from Vesuvius, formed in a similar temperature range (500–800°C): vesuvianite (Balassone *et al.*, 2011). The five-coordinated site in vesuvianite Y1 can be occupied predominately by Fe²⁺, Fe³⁺, Mn²⁺, Mn³⁺, Cu, Mg and Al (Groat *et al.*, 1992; Ohkawa *et al.*, 1992; Armbruster *et al.*, 2002; Panikorovskii *et al.*, 2017a, 2017b, 2017c; 2017d). The unit-cell volume of the litidionite-group minerals depends on the coordination number of the M site (Fig. 7b). The unit-cell volume increases together with the coordination number of the M site. Such variability is the key for the new litidionite-group mineral searches and for the synthesis of new compounds with the litidionite structure, Co and Ni analogues having already been synthesised (Durand *et al.*, 1997).

The litidionite-crystal-structure type is unique because it may contain Ca with a coordination of both number 5 and 6, which is extremely rare for inorganic compounds. It seems that the Ca coordination depends on the formation conditions.

During the voting on the new mineral proposal, one comment was related to the possible formation of enricofrancoite through the dehydration of calcinaksite, because the sample has been stored in a museum collection for a long time. In favour of this assertion is a small contraction of the volume between enricofrancoite 472.74 and calcinaksite 495.94 Å³.

The most unstable water-free minerals in fumarolic environments at normal temperatures are typically associated with halides. For example, sanguite, KCuCl₃ in humid air alters to eriochalcite CuCl₂ × 2H₂O and sylvite KCl after several weeks (Pekov *et al.*, 2015). Another example is saranchinaite, Na₂Cu(SO₄)₂, which transforms into kröhnkite, Na₂Cu(SO₄)₂ × 2H₂O, after one week if exposed at ambient conditions (87% humidity and 25°C) (Siidra *et al.*, 2018). At the same time, the dehydration process is also possible; for example, kröhnkite is replaced by saranchinaite if heated to a temperature above 200°C (Hawthorne and Ferguson, 1975). Silicate minerals are more stable and do not usually change in chemical composition, for example analcime,

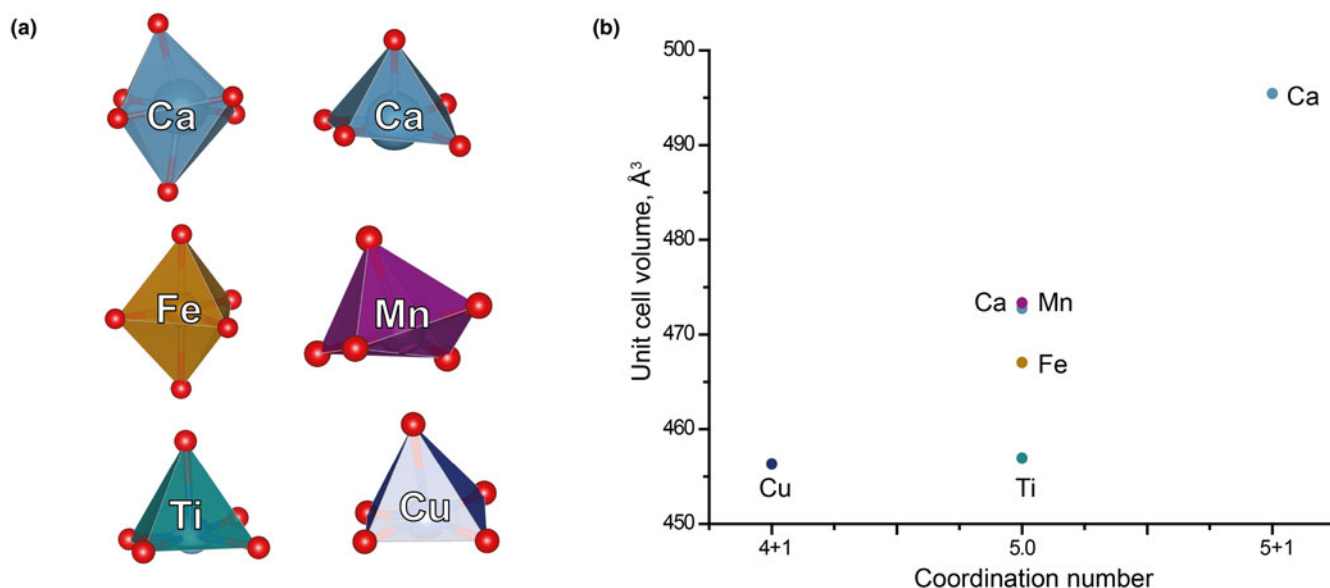


Figure 7. (a) Coordination polyhedra in the minerals of litidionite group based on the data of Rozhdestvenskaya *et al.* (2004), Karimova and Burns (2008), Brandão *et al.* (2009), Chukanov *et al.*, 2015) and Balassone *et al.* (2022); (b) graph of the dependence of unit cell volume on coordination number for litidionite-group minerals.

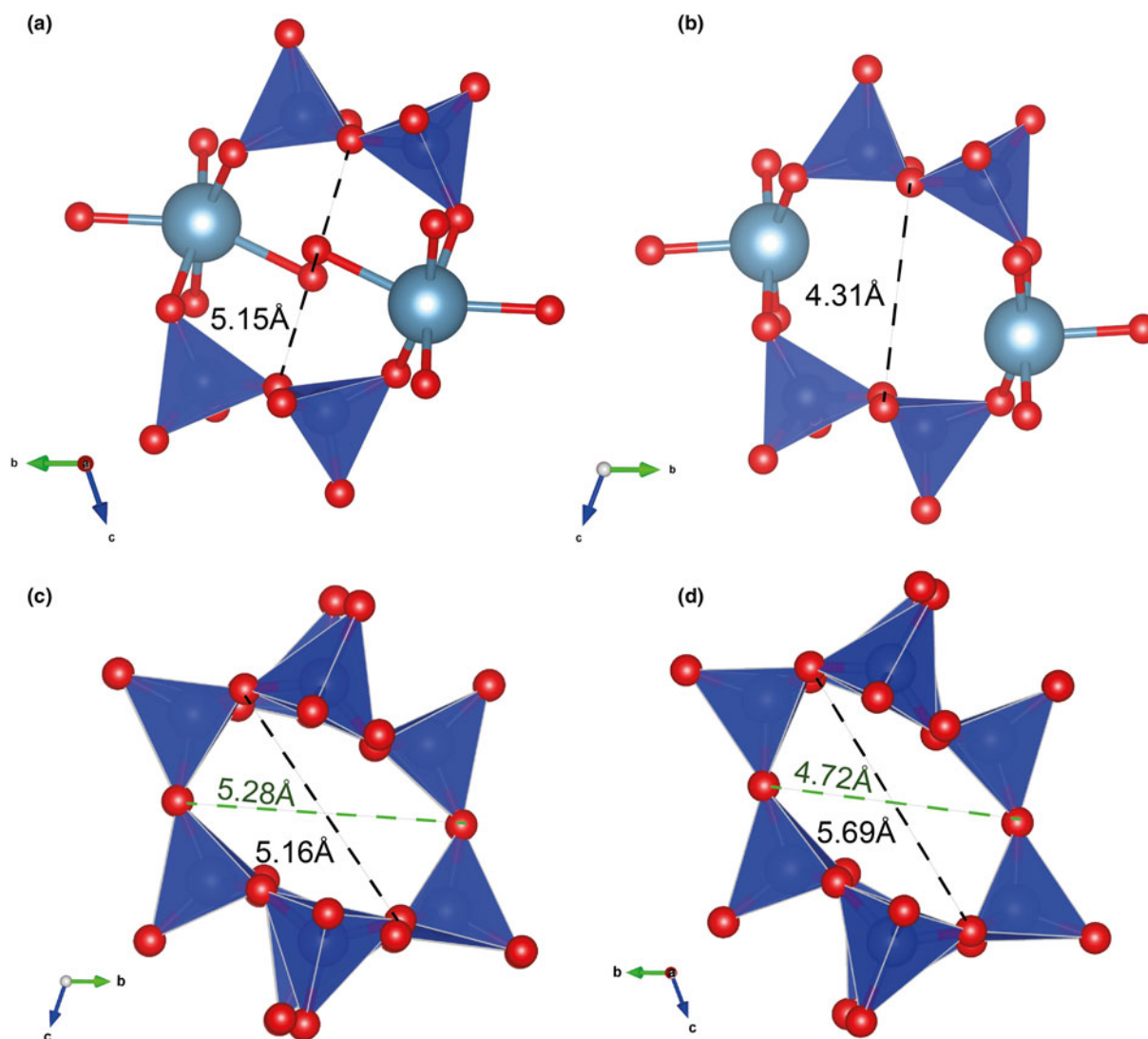


Figure 8. (a) Local coordination of Na-containing channel along the [110] direction in the crystal structure of calcanaksite and (b) enricofrancoite; and (c) geometry of the $[\text{Si}_8\text{O}_{20}]^{8-}$ tubes in the crystal structure of calcanaksite and (d) enricofrancoite.

cancrinite, chabazite-K, gonnardite, phillipsite-K, thomsonite-Ca, scolecite from Somma–Vesuvius Complex are not known to lose water during storage in a museum.

Theoretically, enricofrancoite can be transformed into calcanaksite by a reaction: $\text{KNaCaSi}_4\text{O}_{10} + \text{H}_2\text{O} \rightarrow \text{KNaCaSi}_4\text{O}_{10} \times \text{H}_2\text{O}$. Figures 8a,b show the local coordination of the M position within Na-containing channels along the [110] direction in calcanaksite and enricofrancoite, respectively. The presence of an additional H_2O molecule in calcanaksite leads to an increase in channel size, for example, the O4–O4 distance increases from 4.31 in enricofrancoite to 5.15 Å in calcanaksite. The change of the coordination number for the M site significantly distorts the framework and affects the geometry of the Si_8O_{20} tubes. The large diagonal (Fig. 8c,d) decreases from 5.69 to 5.16 Å and the small diagonal increases from 3.27 to 3.41 Å in enricofrancoite compared with calcanaksite. Thus, the hydration reaction entails global changes in the mineral structure, which requires significant energy and this is unlikely at room conditions. At the very least, such a reaction will be connected with crystallinity loss. Thus, we believe that

a spontaneous transition from enricofrancoite to calcanaksite is unlikely, and such a rearrangement of the structure is likely to entail a loss of crystallinity.

Conclusions

Enricofrancoite with litidionite associations of Somma–Vesuvius are typically found in deep blue glassy crusts and characterise very unusual thermally modified pyroclastic fragments related to the old fumarolic activity of the 1872 eruption (Scacchi, 1880; Pozas *et al.*, 1975). Whereas calcanaksite is a product of contact metamorphism formed during the relatively high-temperature hydrothermal stage (estimated temperature of crystallisation is 300°C), the enricofrancoite can be formed in the temperature range 600–800°C (hydrogen-free), as a result of direct deposition from the gas phase (as volcanic sublimes) or gas–rock interactions. The occurrence of coexisting litidionite-group minerals (earlier litidionite and later ‘Ti-litidionite’), which do not contain OH and H_2O in their composition, corroborates this observation. The

formation of enricofrancoite as a product of dehydration of calcinaksite is possible, but requires significant structural changes and probably leads to loss of crystallinity.

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Competing interests. The authors declare none.

References

- Agilent Technologies (2014) *CrysAlis CCD and CrysAlis RED*. Oxford Diffraction Ltd, Yarnton, Oxfordshire.
- Aksenov S.M., Rastsvetaeva R.K., Chukanov N. V. and Kolitsch U. (2014) Structure of calcinaksite $\text{KNa}[\text{Ca}(\text{H}_2\text{O})][\text{Si}_4\text{O}_{10}]$, the first hydrous member of the litidionite group of silicates with $[\text{Si}_8\text{O}_{20}]_8^-$ tubes. *Acta Crystallographica*, **B70**, 768–775.
- Armbruster T., Gnos E., Dixon R., Gutzmer J., Hejny C., Döbelin N. and Medenbach O. (2002) Manganvesuvianite and tweddillite, two new Mn³⁺-silicate minerals from the Kalahari manganese fields, South Africa. *Mineralogical Magazine*, **66**, 121–135.
- Balassone G., Talla D., Beran A., Mormone A., Altomare A., Moliterni A., Mondillo N., Saviano M. and Petti C. (2011) Vesuvianite from Somma-Vesuvius volcano (southern Italy): Chemical, X-ray diffraction and single-crystal polarized FTIR investigations. *Periodico di Mineralogia*, **80**, 369–384.
- Balassone G., Petti C., Mondillo N., Panikorovskii T.L., de Gennaro R., Cappelletti P., Altomare A., Corriero N., Cangiano M. and D’Orazio L. (2019) Copper Minerals at Vesuvius Volcano (Southern Italy): A Mineralogical Review. *Minerals*, **9**, 730.
- Balassone G., Panikorovskii T.L., Pellino A., Bazai A.V., Bocharov V.N., Krivovichev S.V., Petti C., Cappelletti P. and Mondillo N. (2022) The complex mechanism of Ti^{4+} incorporation into litidionite from the Somma-Vesuvius volcano, Italy. *Mineralogical Magazine*, **86**, 222–233.
- Balassone P., Panikorovskii T.L., Pellino A., Bazai A.V., Bocharov V.N., Goychuk O.F., Avdontseva E.Y., Yakovenchuk V.N., Krivovichev S.V., Petti C., Cappelletti P.A., Mondillo N., Moliterni A. and Altomare A. (2023) Enricofrancoite, IMA 2023-002. CNMNC Newsletter 74. *Mineralogical Magazine*, **87**, 783–787. <https://doi.org/10.1180/mgm.2023.54>.
- Baur W.H. (1974) The geometry of polyhedral distortions. Predictive relationships for the phosphate group. *Acta Crystallographica*, **B30**, 1195–1215.
- Brandão P., Rocha J., Reis M.S., dos Santos A.M. and Jin R. (2009) Magnetic properties of compounds. *Journal of Solid State Chemistry*, **182**, 253–258.
- Breese N.E. and O’Keefe M. (1991) Bond-valence parameters for solids. *Acta Crystallographica*, **B47**, 192–197.
- Chukanov N.V., Aksenov S.M., Rastsvetaeva R.K., Blass G., Varlamov D.A., Pekov I.V., Belakovskiy D.I. and Gurzhiy V.V. (2015) Calcinaksite, $\text{KNaCa}(\text{Si}_4\text{O}_{10})\text{H}_2\text{O}$, a new mineral from the Eifel volcanic area, Germany. *Mineralogy and Petrology*, **109**, 397–404.
- Day M.C. and Hawthorne F.C. (2020) A structure hierarchy for silicate minerals: chain, ribbon, and tube silicates. *Mineralogical Magazine*, **84**, 165–244.
- de Gennaro M. and Franco E. (1974) La K-Cabasite di alcuni «tufi del Vesuvio». *Atti della Accademia Nazionale dei Lincei*, **60**, 490–497.
- Dolivo-Dobrovolsky D.D. (2016) *MINAL*, free software. Saint-Petersburg, Russia. <<http://www.dimadd.ru>>.
- Dolomanov O. V., Bourhis L.J., Gildea R.J., Howard J.A.K. and Puschmann H. (2009) OLEX2 : a complete structure solution, refinement and analysis program. *Journal of Applied Crystallography*, **42**, 339–341.
- Durand G., Vilminot S., Richard-Plouet M., Derory A., Lambour J.P. and Drillon M. (1997) Magnetic behavior of $\text{Na}_2\text{MSi}_4\text{O}_{10}$ (M=Co, Ni) compounds. *Journal of Solid State Chemistry*, **131**, 335–340.
- Franco E. and de Gennaro M. (1988) Panunzite; a new mineral from Mt. Somma-Vesuvio, Italy. *American Mineralogist*, **73**, 420–421.
- Galuskin E. V., Lazic B., Armbruster T., Galuskina I.O., Pertsev N.N., Gazeev V.M., Wlodyka R., Dulski M., Dzierzanowski P., Zadov A.E. and Dubrovinsky L.S. (2012) Edgrewite $\text{Ca}_9(\text{SiO}_4)_4\text{F}_2$ -hydroxyledgrewite $\text{Ca}_9(\text{SiO}_4)_4(\text{OH})_2$, a new series of calcium humite-group minerals from altered xenoliths in the ignimbrite of Upper Chegem caldera, Northern Caucasus, Kabardino-Balkaria, Russia. *American Mineralogist*, **97**, 1998–2006.
- Golovachev V.P., Drozdov Y.N., Kuz’min E.A. and Belov N.V. (1970) The crystal structure of fenaksite, $\text{NaKFeSi}_4\text{O}_{10}$. *Doklady Akademii Nauk SSSR*, **194**, 818–820 [in Russian].
- Groat L.A., Hawthorne F.C. and Ercit T.S. (1992) The chemistry of Vesuvianite. *The Canadian Mineralogist*, **30**, 19–48.
- Hawthorne F.C. and Ferguson R.B. (1975) Refinement of the crystal structure of kröhnkite. *Acta Crystallographica*, **B31**, 1753–1755.
- Holland T.J.B. and Redfern S.A.T. (1997) Unit cell refinement from powder diffraction data: the use of regression diagnostics. *Mineralogical Magazine*, **61**, 65–77.
- Karimova O. and Burns P.C. (2008) Silicate tubes in the crystal structure of manaksite. Pp. 153–156 in: *Minerals as Advanced Materials I*. Springer Berlin Heidelberg, Berlin, Heidelberg.
- Katz A.K., Glusker J.P., Beebe S.A. and Bock C.W. (1996) Calcium ion coordination: a comparison with that of beryllium, magnesium, and zinc. *Journal of the American Chemical Society*, **118**, 5752–5763.
- Khomyakov A.P., Kurova T.A. and Nechelyustov G.N. (1992) Manaksite $\text{NaKMnSi}_4\text{O}_{10}$: a new mineral. *Zapiski RMO*, **121**, 112–114.
- Kornev A.N., Maksimov B.A., Lider V.V., Ilyukhin V.V. and Belov N.V. (1972) Crystal structure of $\text{Na}_2\text{CuSi}_4\text{O}_{10}$. *Soviet Physics Doklady*, **17**, 735–737.
- Lafuente B., Downs R.T., Yang H. and Stone N. (2015) The power of databases: the RRUFF project. Pp. 1–30 in: *Highlights in Mineralogical Crystallography* (T. Armbruster and R.M. Danisi, editors). W. De Gruyter, Berlin, Germany.
- Liebau F. (1985) *Structural Chemistry of Silicates*. Springer, Berlin, Heidelberg.
- Mandarino J.A. (2007) The Gladstone-Dale compatibility of minerals and its use in selecting mineral species for further study. *The Canadian Mineralogist*, **45**, 1307–1324.
- McCusker L.B., Liebau F. and Engelhardt G. (2003) Nomenclature of structural and compositional characteristics of ordered microporous and mesoporous materials with inorganic hosts. *Microporous and Mesoporous Materials*, **58**, 3–13.
- Merlino S., Franco E., Mattia C.A., Pasero M. and De Gennaro M. (1985) The crystal structure of panunzite (natural tetrakalsilite). *Neues Jahrbuch Fur Mineralogie Monatshefte*, **7**, 322–328.
- Momma K. and Izumi F. (2011) VESTA 3 for three-dimensional visualization of crystal, volumetric and morphology data. *Journal of Applied Crystallography*, **44**, 1272–1276.
- Ohkawa M., Yoshiasa A. and Takeno S. (1992) Crystal chemistry of vesuvianite: site preferences of square- pyramidal coordinated sites. *American Mineralogist*, **77**, 945–953.
- Pakhomovsky Y.A., Panikorovskii T.L., Yakovenchuk V.N., Ivanyuk G.Yu., Mikhailova J.A., Krivovichev S.V., Bocharov V.N. and Kalashnikov A.O. (2018) Selivanovaite, $\text{NaTi}_3(\text{Ti},\text{Na},\text{Fe},\text{Mn})_4[(\text{Si}_2\text{O}_7)_2\text{O}_4(\text{OH},\text{H}_2\text{O})_4]\cdot n\text{H}_2\text{O}$, a new rock-forming mineral from the eudialyte-rich malignite of the Lovozero alkaline massif (Kola Peninsula, Russia). *European Journal of Mineralogy*, **30**, 525–535.
- Panikorovskii T.L., Chukanov N.V., Aksenov S.M., Mazur A.S., Avdontseva E.Yu., Shilovskikh V.V. and Krivovichev S.V. (2017a) Alumovesuvianite, $\text{Ca}_{19}\text{Al}(\text{Al},\text{Mg})_{12}\text{Si}_{18}\text{O}_{69}(\text{OH})_9$, a new vesuvianite-group member from the Jeffrey mine, asbestos, Estrie region, Québec, Canada. *Mineralogy and Petrology*, **111**, 833–842.
- Panikorovskii T.L., Chukanov N.V., Rusakov V.S., Shilovskikh V.V., Mazur A.S., Balassone G., Ivanyuk G.Yu. and Krivovichev S.V. (2017b) Vesuvianite from the Somma-Vesuvius Complex: New Data and Revised Formula. *Minerals*, **7**, 248.

- Panikorovskii T.L., Shilovskikh V.V., Avdontseva E.Yu., Zolotarev A.A., Karpenko V.Yu., Mazur A.S., Yakovenchuk V.N., Bazai A.V., Krivovichev S.V. and Pekov I.V. (2017c) Magnesiovesuvianite, $\text{Ca}_{19}\text{Mg}(\text{Al},\text{Mg})_{12}\text{Si}_{18}\text{O}_{69}(\text{OH})_9$, a new vesuvianite-group mineral. *Journal of Geosciences (Czech Republic)*, **62**, 25–36.
- Panikorovskii T.L., Shilovskikh V.V., Avdontseva E.Yu., Zolotarev A.A., Pekov I.V., Britvin S.N., Hälenius U. and Krivovichev S.V. (2017d) Cyprine, $\text{Ca}_{19}\text{Cu}^{2+}(\text{Al},\text{Mg},\text{Mn})_{12}\text{Si}_{18}\text{O}_{69}(\text{OH})_9$, a new vesuvianite-group mineral from the Wessels mine, South Africa. *European Journal of Mineralogy*, **29**, 295–306.
- Pekov I.V., Zubkova N.V., Belakovskiy D.I., Lykova I.S., Yapaskurt V.O., Viggasina M.F., Sidorov E.G. and Pushcharovsky D.Yu. (2015) Sanguite, KCuCl_3 , A new mineral from the Tolbachik Volcano, Kamchatka, Russia. *The Canadian Mineralogist*, **53**, 633–641.
- Pozas J.M., Rossi G. and Tazzoli V. (1975) Re-examination and crystal structure analysis of litidionite. *American Mineralogist*, **60**, 471–474.
- Rainho J.P., Carlos L.D. and Rocha J. (2000) New phosphors based on Eu^{3+} -doped microporous titanosilicates. *Journal of Luminescence*, **87–89**, 1083–1086.
- Rozhdestvenskaya I.V., Bannova I.I., Nikishova L.V. and Soboleva T.V. (2004) The crystal structure of fenaksite $\text{K}_2\text{Na}_2\text{Fe}_2\text{Si}_8\text{O}_{20}$. *Doklady of the Russian Academy of Sciences*, **398**, 1029–1033.
- Scacchi E. (1880) Lapilli azzurri del Vesuvio. *Rendiconto dell'Accademia delle Scienze Fisiche e Matematiche*, **19**, 175–179.
- Sheldrick G.M. (2015) Crystal structure refinement with SHELXL. *Acta Crystallographica*, **C71**, 3–8.
- Siidra O.I., Lukina E.A., Nazarchuk E.V., Depmeier W., Bubnova R.S., Agakhanov A.A., Avdontseva E.Yu., Filatov S.K. and Kovrugin V.M. (2018) Saranchinaite, $\text{Na}_2\text{Cu}(\text{SO}_4)_2$, a new exhalative mineral from Tolbachik volcano, Kamchatka, Russia, and a product of the reversible dehydration of kröhnkite, $\text{Na}_2\text{Cu}(\text{SO}_4)_2(\text{H}_2\text{O})_2$. *Mineralogical Magazine*, **82**, 257–274.
- Yakovenchuk V., Pakhomovsky Y., Panikorovskii T., Zolotarev A., Mikhailova J., Bocharov V., Krivovichev S. and Ivanyuk G. (2019) Chirvinskyite, $(\text{Na}, \text{Ca})_{13}(\text{Fe}, \text{Mn}, \square)_2(\text{Ti}, \text{Nb})_2(\text{Zr}, \text{Ti})^{3-}(\text{Si}_2\text{O}_7)_4(\text{OH}, \text{O}, \text{F})_{12}$, a new mineral with a modular wallpaper structure, from the Khibiny Alkaline Massif (Kola Peninsula, Russia). *Minerals*, **9**, 219.