

Dynamic Oxygen Motion in Irradiated-Annealed High Temperature Superconducting Wire

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Research over the past years has shown that proton and ion irradiation can significantly improve pinning in second-generation (2G) HTS wire by producing highly engineered defects [1-3]. However, the enhancement in J_c usually comes with a cost of depression of critical temperature T_c . The degradation can be partially recovered by post-annealing process [4]. The film quality and superconducting properties can be improved by altering the oxygen disorder and nanoscale defects after annealing. Although irradiation defects have been studied over a full range of atomic, electronic structures and physical properties, a detailed atomic level TEM analysis on the annealing effects is still needed.

Here, we use the atomically resolved electron energy loss spectroscopy (EELS) to locate the oxygen vacancies in the planes and chains of YBCO. A 1.2 μm thick MOD REBCO-based 2G wire was irradiated with 18-MeV Au⁵⁺ ions to a dose of 2×10^{11} Au/cm² along the c -axis of YBCO film [5]. Prior to the irradiation, the wire sample has a 77 K, self-field I_c of around 385 A/cm-w. In comparison, the value decreased to 320 A/cm-w after irradiation. The irradiated samples were then annealed in oxygen atmosphere at various temperatures, which led to an enhancement in 77 K, self-field I_c to around 358 A/cm-w.

We present a detailed investigation of the oxygen behavior during the irradiation and annealing processes by probing electronic structure of different sites in YBCO matrix. We extracted the Cu- $L_{2,3}$ edge spectra from different layers of YBCO structure in the non-irradiated, as-irradiated, and annealed samples to clarify the location of oxygen vacancies. The Cu L_3 and L_2 white lines are very sensitive to the change in valence states. The chain layer spectrum in the non-irradiated film (Figure 1a) shows splitting peaks at 934.5 eV (peak a) and 937.2 eV (peak b), due to the existence of mixed valence states of Cu²⁺ (peak a) and Cu¹⁺ (peak b). The plane layer shows only peak a which is consistent with the previous reported YBCO₇ data [6]. In the as-irradiated film (Figure 1b), the chain layer shows a strong main peak b with a weak shoulder peak a , which corresponds to predominant Cu¹⁺ ions in Cu-O chains. The CuO₂ plane layer shows splitting peaks a and b , which indicates the coexistence of Cu²⁺ and Cu¹⁺ oxidation states. The previous research showed that plane layer Cu- $L_{2,3}$ does not show a significant change in YBCO_{7- δ} from $\delta=1$ (full chains) to $\delta=0$ (empty chains). The splitting plane layer spectrum in the as-irradiated film can be caused by the oxygen displacement in both planes and chains. After annealing in oxygen atmosphere (Figure 1c), the chain layers still show main peak b with a shoulder peak a . However, the plane layers have the similar Cu edges as non-irradiated sample. The results indicate the O doping during annealing process is preferred to reoccupy the O vacancy sites in plane layers.

In conclusion, we performed a detailed analysis of the reversible dynamic behavior of oxygen vacancies in YBCO film during irradiation-annealing processes. The low energy heavy ion irradiation creates oxygen point defects in both chain and plane layer of YBCO film. The plane sites oxygen vacancies were recovered in post-annealing process. The suppressed T_c and self-field I_c in the irradiated samples were partially recovered after annealing. The variation trend is consistent with the dynamic behavior of oxygen defects, which indicates the strong influence of plane site oxygen vacancies on the superconductivity properties in the irradiated HTS wires [7].

References:

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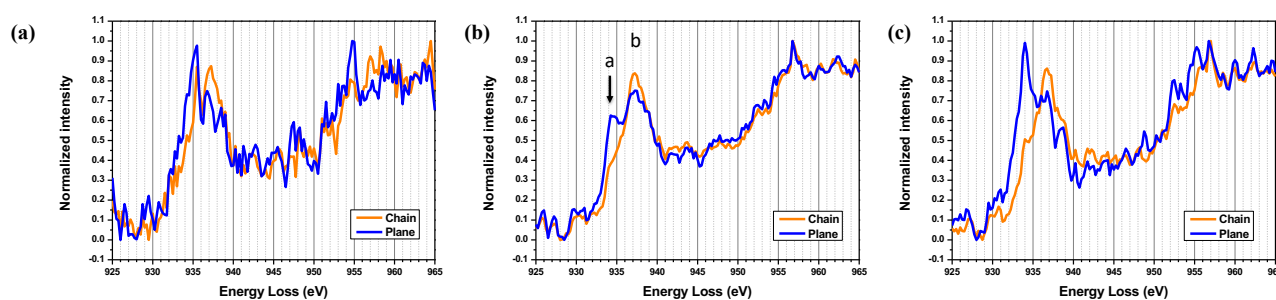


Figure 1. Cu- $L_{2,3}$ edge spectra are extracted from different layers of YBCO in (a) non-irradiated, (b) as-irradiated, and (c) annealed films. In non-irradiated film, the spectra show splitting Cu- L_3 edge in chain layer and peak a in plane layer, which is consistent with the YBCO₇ structure. In (b), the chain layer shows one peak, which corresponds to Cu¹⁺ ions in Cu-O chains. Cu L_3 edge of CuO₂ plane layer shows splitting peaks a and b , indicating the coexistence of Cu²⁺ and Cu¹⁺ oxidation states. (c) After annealing, the chain layers still show peak b , but the plane layers have only one strong peak a .