

THE CHEMISTRY OF COOL CIRCUMSTELLAR ENVELOPES

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ABSTRACT. We have developed a time-dependent chemical kinetic model to describe the chemistry in the circumstellar envelopes of cool stars, with particular reference to IRC + 10216. Our detailed calculations show that ion-molecule reactions are important in the formation of many of the species observed in IRC + 10216.

1. INTRODUCTION

The envelope of IRC + 10216 is observed to contain many complex molecules nearly all of which have also been detected in the dark dust cloud, TMC-1. In this source, cosmic-ray ionisation drives the chemistry which, at least for the smaller molecules, is fairly well understood (Millar and Freeman 1984, Leung, Herbst and Huebner 1984). Ion-molecule chemistry can also occur in IRC + 10216 driven by both cosmic-rays and the external interstellar ultraviolet radiation field (Huggins and Glassgold 1982, Nejad, Millar and Freeman 1984). Parent species injected into the envelope at $r_0 = 10^{16}$ cm are CO, C₂H₂, HCN, CH₄, NH₃ and N₂, though this latter species has only a minor effect on the chemistry. These parents give rise to a system which totals 80 species whose radial abundances are calculated by integrating the resulting system of stiff differential chemical kinetic equations through use of the GEAR method. In Table 1 we present a summary of one of the many calculations performed by us.

2. DISCUSSION

The absorption of the external UV field by dust grains in the envelope ensures that at $r = 10^{16}$ cm the chemistry is driven by cosmic-ray ionisation of H₂. The H₃⁺ ion so produced has a low proton affinity and undergoes proton transfer reactions with CO, C₂H₂, HCN and NH₃. The ion HCO⁺ does not have a large abundance however because it transfers protons to C₂H₂, HCN and NH₃. The formation of HCNH⁺ can lead to the production of HNC through dissociative recombination while

TABLE 1. Calculated column densities, $N(\text{cm}^{-2})$, for IRC + 10216 using $\dot{M} = 5 \times 10^{-5} M_{\odot} \text{yr}^{-1}$ and $V_e = 16 \text{ km s}^{-1}$.

Species	N	Species	N	Species	N
CN	7.4(14)	C ₂ H	1.8(15)	HCO ⁺	1.7(12)
HCN	2.1(13)	C ₃ H	5.5(13)	HCNH ⁺	1.8(12)
CCN	4.4(11)	C ₃ H ₂	5.7(13)	CH ₂ CO	6.0(11)
C ₃ N	5.2(12)	C ₄ H	2.1(14)	CH ₃ CN	1.0(13)
HC ₃ N	4.6(12)	C ₃ O	3.7(12)	C ₂ H ₃	1.1(14)

C₂H₃⁺ reacts with C₂H₂, CH₄ and CO to form more complex species. As one proceeds outward in the envelope, the external UV dominates cosmic-rays and provides a source of ionisation as well as the creation of daughter species. In this region the chemistry is particularly interesting. The most important ions in driving the chemistry become C₂H₂⁺, CH₃⁺ and C⁺. The ion C₂H₂⁺ reacts with many species and leads to C₃O, HC₃N, C₃N, C₄H, C₃H₂, C₃H and CCN. CH₃⁺ reacts slowly with H₂ but has an extremely rapid radiative association with HCN and an association with CO which produce CH₃CN and CH₂CO respectively.

Our detailed calculations enable us to make the following conclusions about ion-molecule chemistry in IRC + 10216.

1. Cosmic-rays and UV radiation provide a source of ionisation which drives an extensive ion-molecule chemistry in the envelope.
2. Given the uncertainties in both the observationally derived column densities and those calculated theoretically, we find a good agreement between observation and theory for many species including CN, C₂H, HNC, C₃H, C₃H₂, C₄H and CH₃CN.
3. The calculated abundances of C₃N and HC₃N are too low by more than an order of magnitude which may imply that they, or at least HC₃N, are formed in the warm, denser gas interior to 10¹⁶ cm.
4. Certain oxygen-bearing species such as C₃O, CH₂CO and HCO⁺ may be detectable in IRC + 10216.
5. Similarities between the chemical composition of TMC-1 and IRC + 10216 can be explained by the occurrence of a similar ion-molecule chemistry in both objects.
6. Differences in the absolute abundances of certain species in these two objects relate to the fact that in IRC + 10216, the chemistry involves the breakdown of stable parent molecules into atoms and atomic ions, while in TMC-1 it involves the build-up of these parents from atoms.

3. REFERENCES

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