Analytical STEM/EDS Characterization of Elemental Segregation and Solid Solution Formation in Multiphase Ceramics

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Bulk ceramic processing includes the densification of powders at high temperature. This study evaluates elemental segregation at grain boundaries and solid solution formation for a three-phase ceramic Al₂O₃-MgAl₂O₄- 8 mole% Y₂O₃ stabilized cubic ZrO₂ (YSZ) made under different processing conditions: (1) conventional sintering, (2) two step-sintering, (3) spark plasma sintering (SPS), and (4) flash sintering [1,2]. In addition, the stability of interfaces and phases under 946 MeV Au ion irradiation is analyzed. Segregation at grain boundaries and point defect generation in the bulk can serve to disrupt phonon propagation, reducing thermal conductivity.

TEM samples were prepared by using a TESCAN GAIA3 FIB-SEM. JEOL 2800 and JEOL Grand ARM300CF TEM/STEM at 300keV, both equipped with dual silicon drift detectors for energy dispersive spectroscopy (EDS), were used to characterize the grain boundaries and to understand elemental segregation. Both EDS spectral maps and line scans were obtained for analysis. The electron beam probe size was 0.1 nm on the Grand ARM and 1 nm on the JEOL 2800. The EDS parameters were optimized based on minimizing beam damage versus increasing net EDS counts.

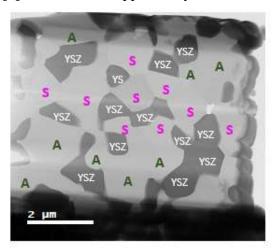
There are three phases, YSZ, alumina, spinel, in this material and six types of grain boundaries (Fig. 1, conventionally sintered). Mg and Al are found to be present in varying amounts in solid solution in YSZ and Al is highly segregated at YSZ/YSZ grain boundaries, with a depletion of Zr (Fig. 2, flash sample). In Spinel/Spinel grain boundaries, there is depletion of Mg at the grain boundary accompanied by the segregation of Y and Zr (Fig. 3, SPS). Similar observations are made for Alumina/Alumina grain boundaries and Spinel/Alumina grain boundaries. In addition, there is a variation of the Mg/Al ratio in the MgAl₂O₄ grains, depending on the sintering process. Higher temperatures reached during processing expand the solution solubility of Al₂O₃ in MgAl₂O₄. YSZ/Spinel and YSZ/Alumina interfaces do not have a significant amount of elemental segregation which is not surprising as these boundaries are heterogeneous with different cations on each side. However, these interfaces in the SPS samples were not as resistant to electron beam damage compared to the conventionally sintered samples. The higher stability in conventionally sintered samples may be due to the structure of the boundaries with high temperature, long anneal times in contrast to the rapid sintering at lower temperatures achieved in SPS resulting in higher energy non-equilibrium grain boundaries.

In SPS samples, $YSZ/MgAl_2O_4$ interfaces show depletion of Mg and O in the spinel grain near the boundary but enrichment of Mg at the interface boundary. All depletion is also observed at YSZ/Al_2O_3 boundaries to some extent. However, All depletion is not as prominent as the loss of Mg and O and is not observed for most boundaries analyzed. The stoichiometric loss of Mg and O is believed to be due to gradual volatilization in the vacuum of TEM [3]. Irradiated SPS samples were found to be even more prone to e-beam damage, particularly at $YSZ/MgAl_2O_4$ grain boundaries (Fig.4, SPS irradiated). The elemental segregation and inhomogeneity observed would all contribute to lower thermal conductivity [4].

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References:

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- [2] M Cologna, ALG Prette and R Raj, J Am Ceram Soc 94 (2011), p. 316.
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- [4] This work is supported by NSF DMR Thermal Grant 442660-21130.



2nm YSZ
Al Zr

Fig. 1: BF STEM image of 3-phase conventionally sintered.

Fig. 2: DF STEM and EDS maps of YSZ/YSZ boundary with Al segregation in flash sintered sample.

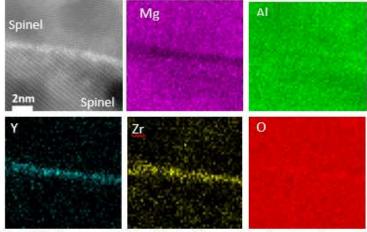


Fig. 3: DF STEM and EDS maps of Spinel/Spinel boundary showing elemental segregation in SPS non-irradiated sample.

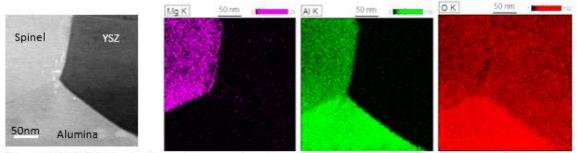


Fig. 4: BF STEM and elemental maps showing e-beam damage after EDS scan in SPS irradiated sample.